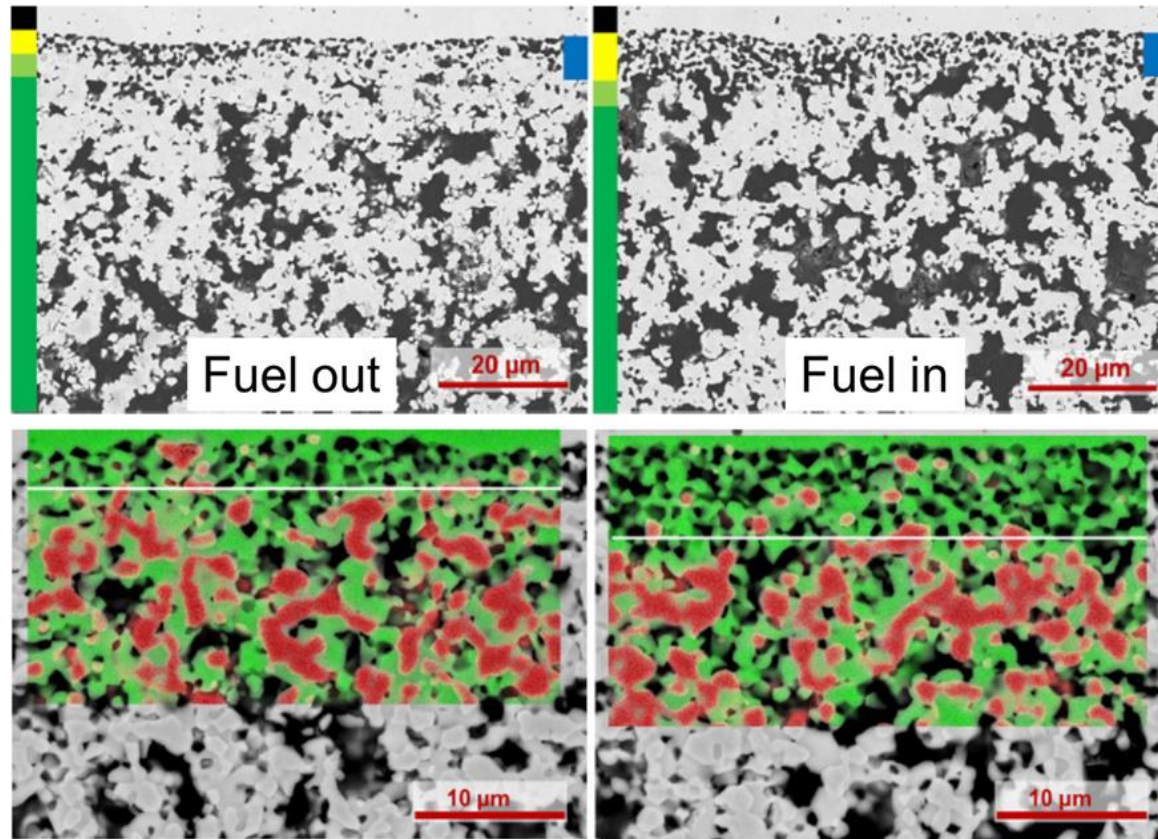


Processing and durability of fuel electrode-supported cells with Ni-GDC electrode

APRIL 16TH, 2026 | CHRISTIAN LENSER, DENISE RAMLER, STEFAN KUCHARSKI, IURII KOGUT, ANDRÉ WEBER,
AND NORBERT H. MENZLER

- (1) Institute of Energy Materials and Devices: Materials Synthesis and Processing (IMD-2), Forschungszentrum Jülich GmbH, Jülich, Germany;
- (2) Institute for Applied Materials (IAM-ET), Karlsruhe Institute of Technology (KIT), Karlsruhe, Germany

Motivation: Ni migration in Ni-YSZ



Ni depletion in fuel electrode after 20.000 h of electrolysis operation

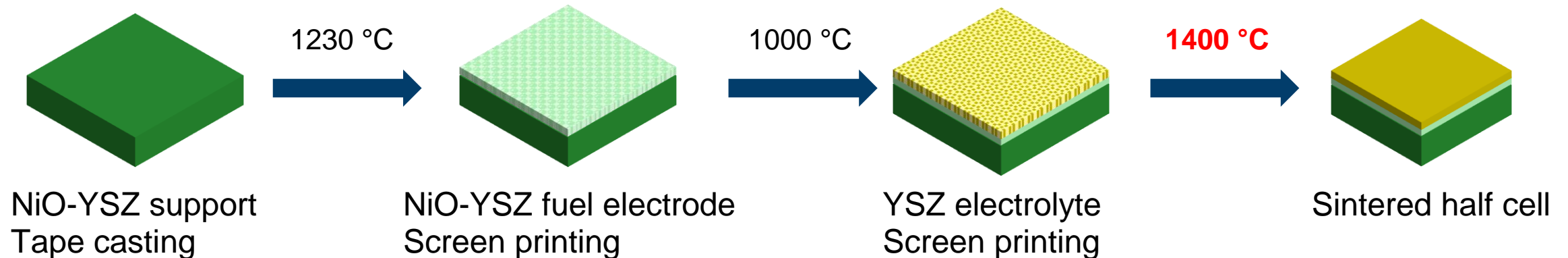
C. Frey, et al., *J. Electrochem. Soc.* (2018);165:F357-F64.

Q. Fang, et al., *J. Electrochem. Soc.* (2018);165:F38-F45.

- Voltage / ASR degradation of 0.6 / 8.2% / kh
- Ni migration as the primary degradation mechanism
- Can Ni-GDC cathodes solve this problem?

Material challenge - Integration of Ni-GDC into FESCs

- Ni-GDC fuel electrode is common in metal-^[1,2] or electrolyte-supported^[3,4] cell designs
- Very rarely encountered in fuel electrode-supported cells
- The problem can be located in the manufacturing route:



[1] D. Udomsilp et al., Cell Reports Physical Science, 1 (2020) 100072.

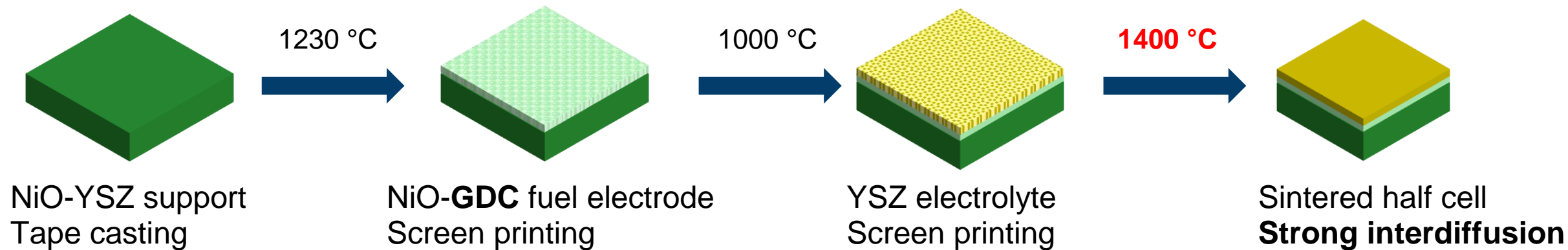
[2] C. Bischof et al., International Journal of Hydrogen Energy, 44 (2019) 31475-31487.

[3] N. Trofimenko et al., ECS Transactions, 78 (2017) 3025-3037.

[4] M. Kusnezoff et al., Materials, 9 (2016) 906.

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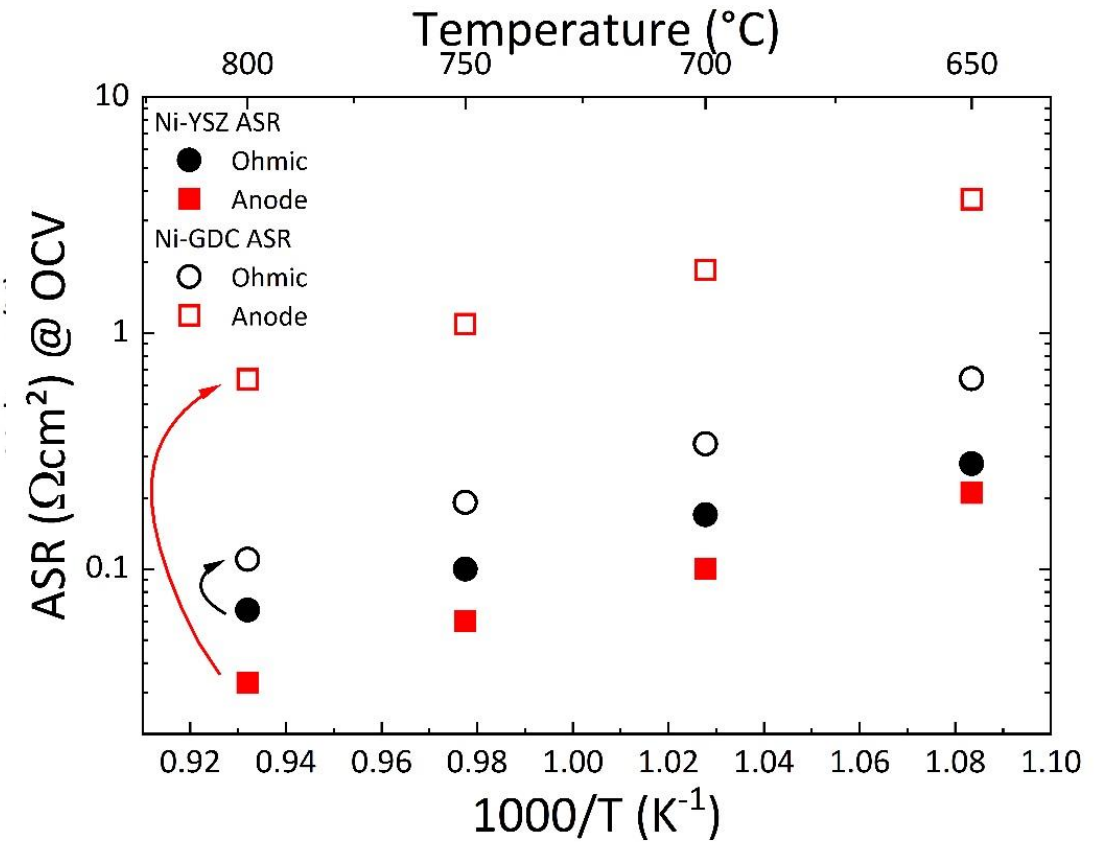
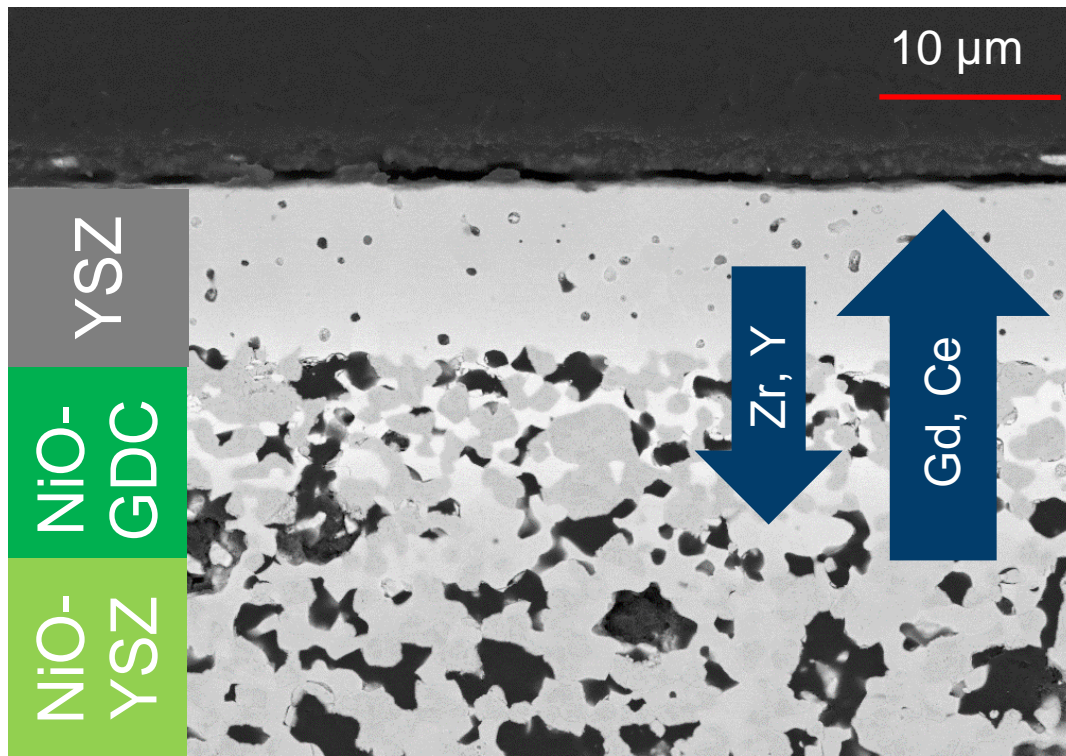
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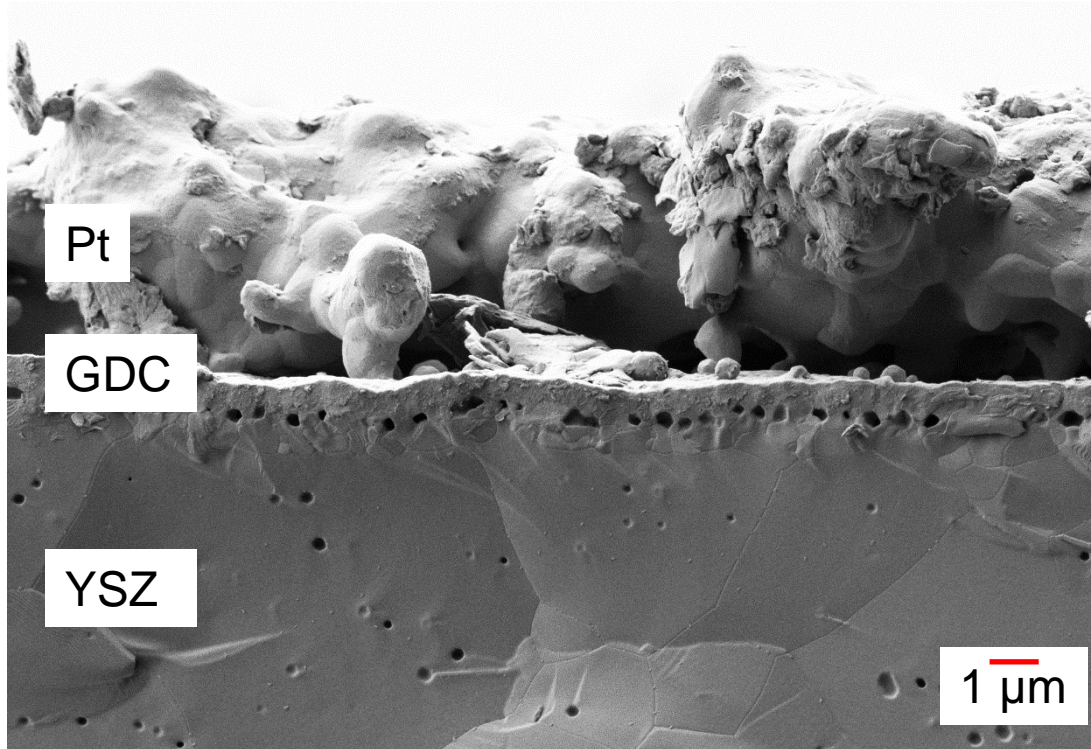
[4] M. Kusnezoff et al., Materials, 9 (2016) 906.

Reduced performance due to interdiffusion

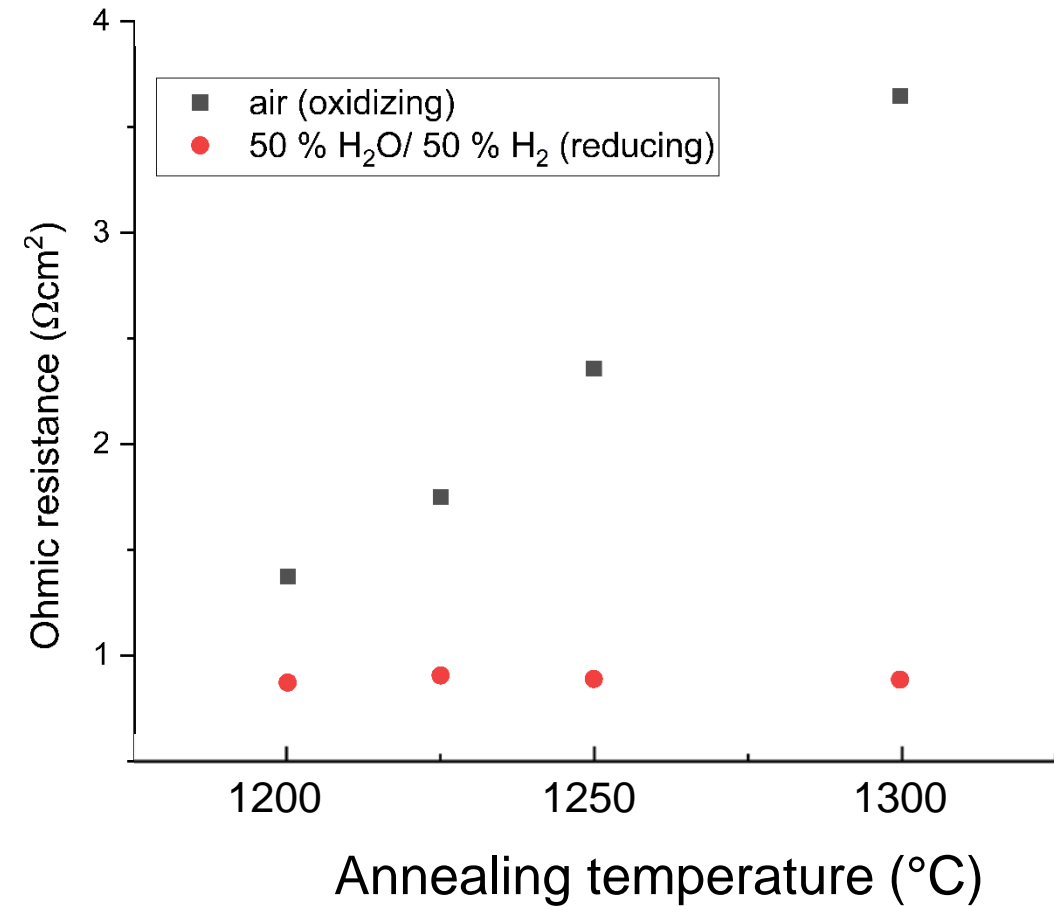
- Substitution of NiO-YSZ by NiO-GDC in FESC manufacturing route → decreased performance
- Well-known: decreasing conductivity, less well-known: Kirkendal porosity



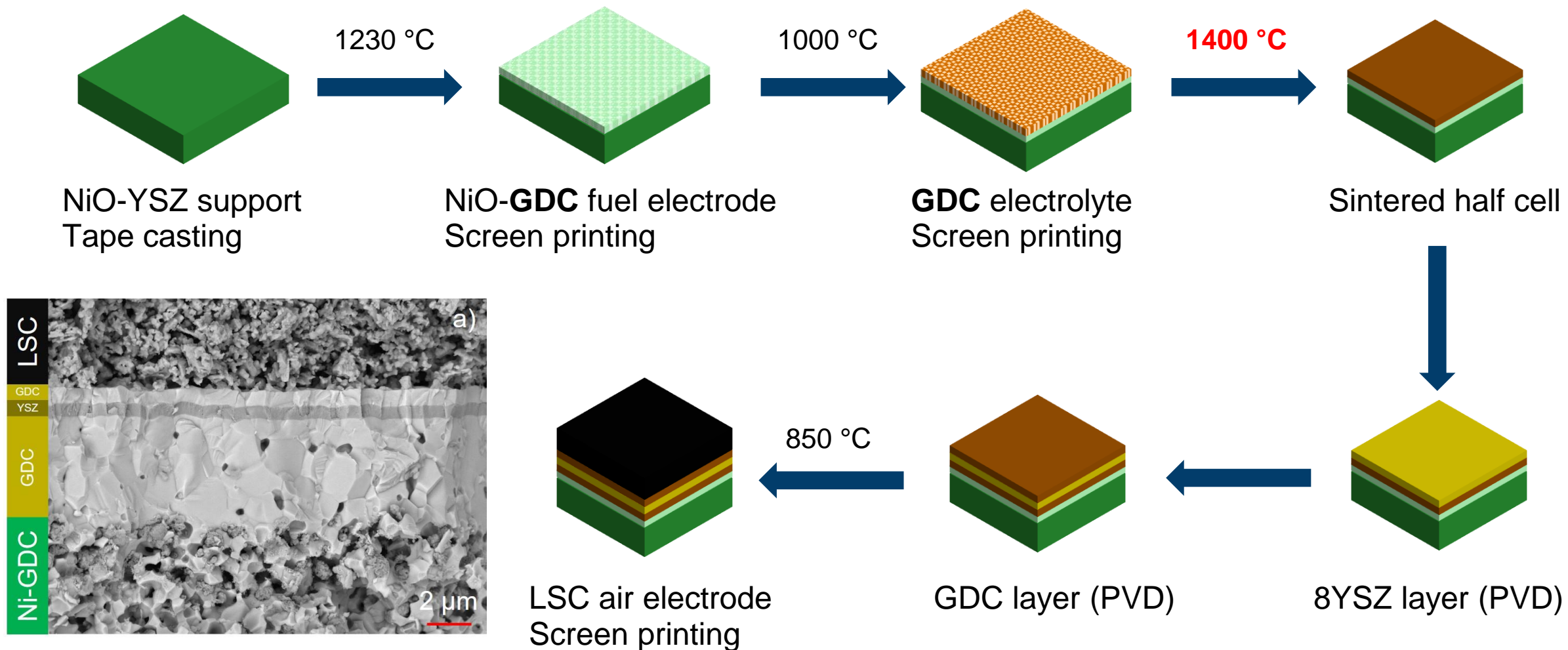
Effect of interdiffusion as a function of pO_2



- Interdiffusion at GDC-YSZ model interface
- R_{ohm} in air, R_{pol} in H_2

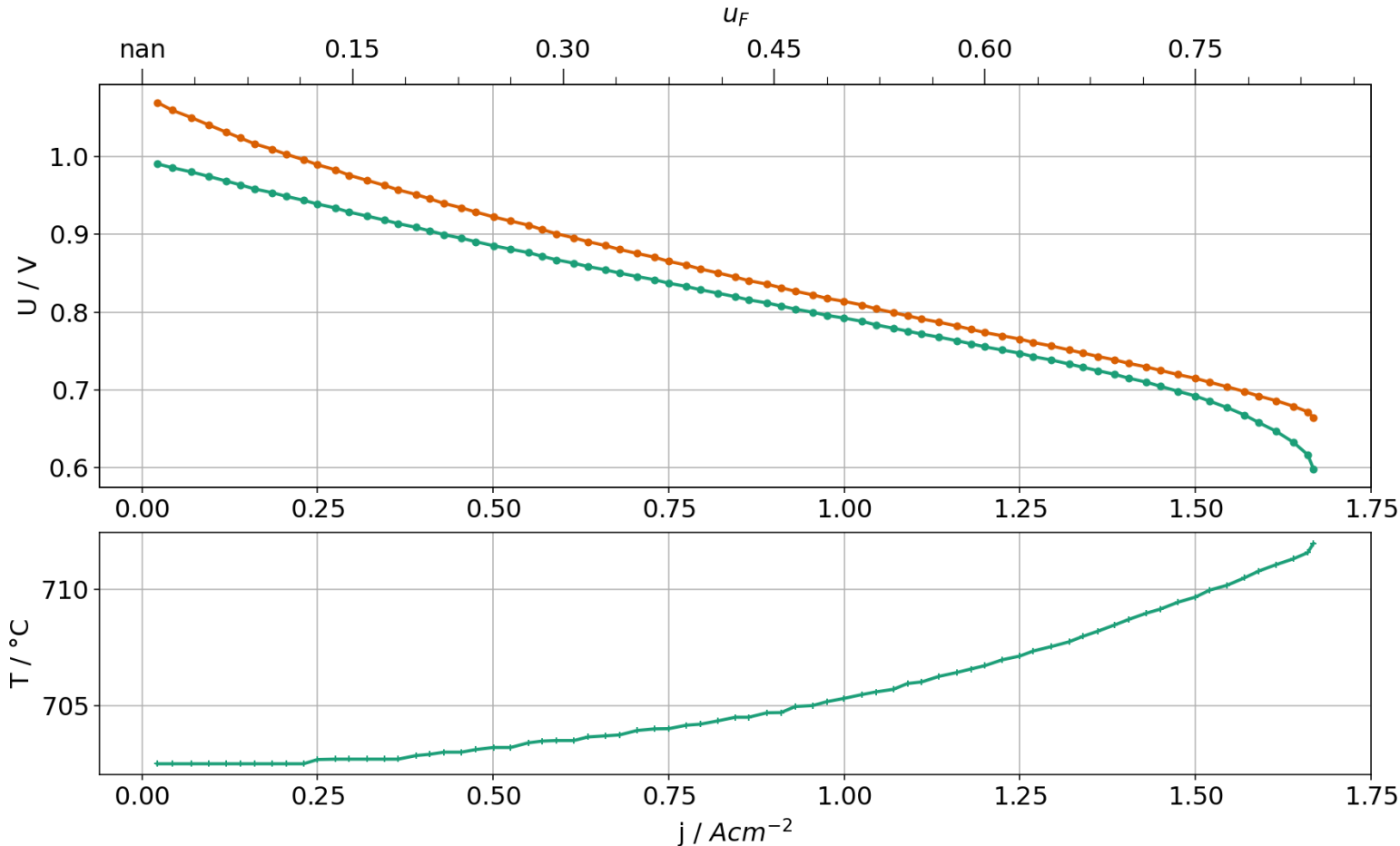


How to avoid interdiffusion



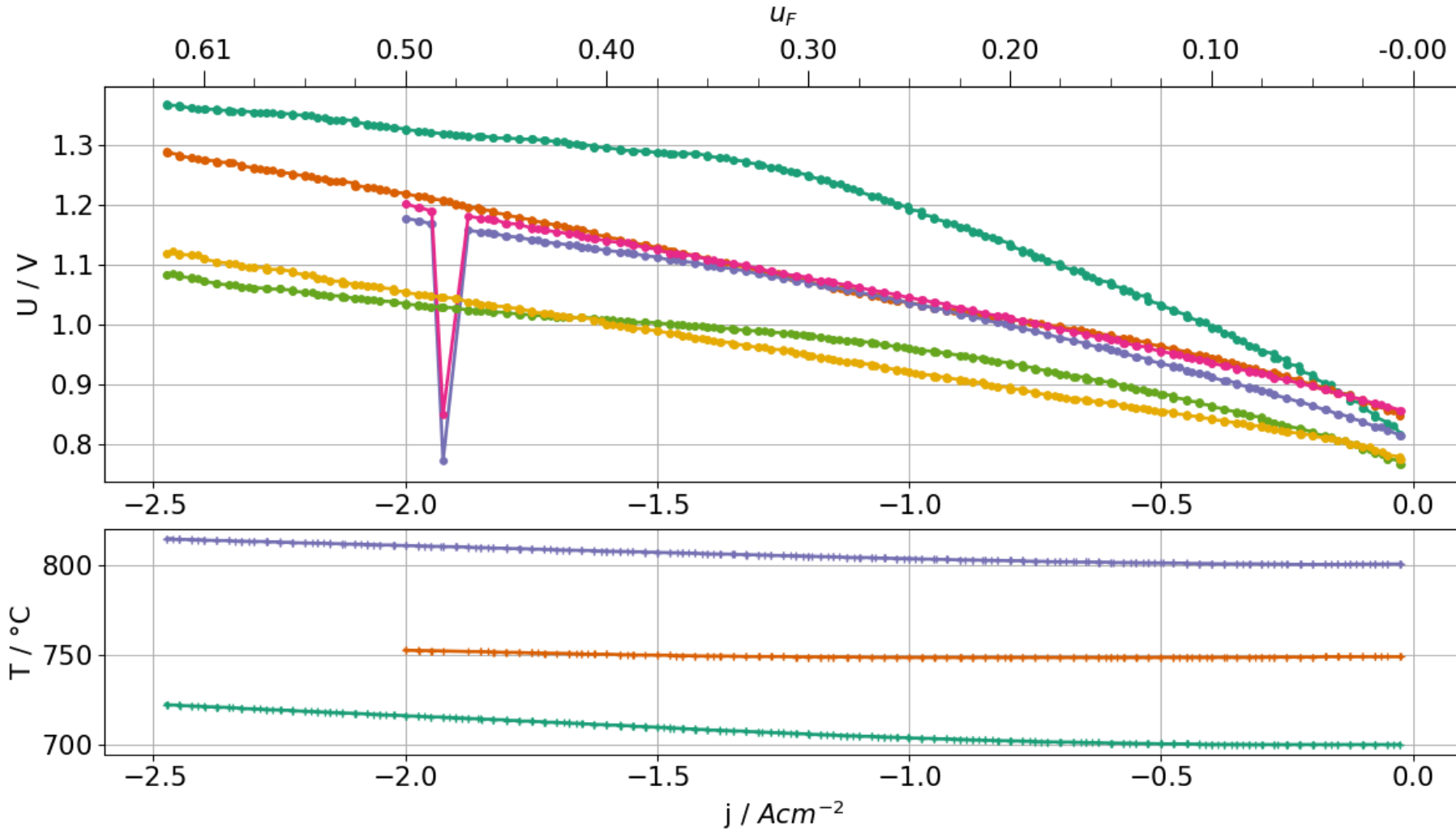
[1] J. Zhang, C. Lenser, N. Russner, A. Weber, N.H. Menzler, O. Guillon, Journal of the American Ceramic Society, 106 (2022) 93-99.

2-layer shortstack with PVD electrolyte



- Dry H_2 vs air
- OCV rather low: 991 mV (layer 1) and 1070 mV (layer 2) at 700°C
- Cell performance is excellent in SOFC mode with $\sim 1.5 \text{ A cm}^{-2}$ at 700 mV (Elcogen: ~ 1.4 ; Type III: ~ 1)
- Early diffusion limitation and low OCV point to gas and/or current leakage

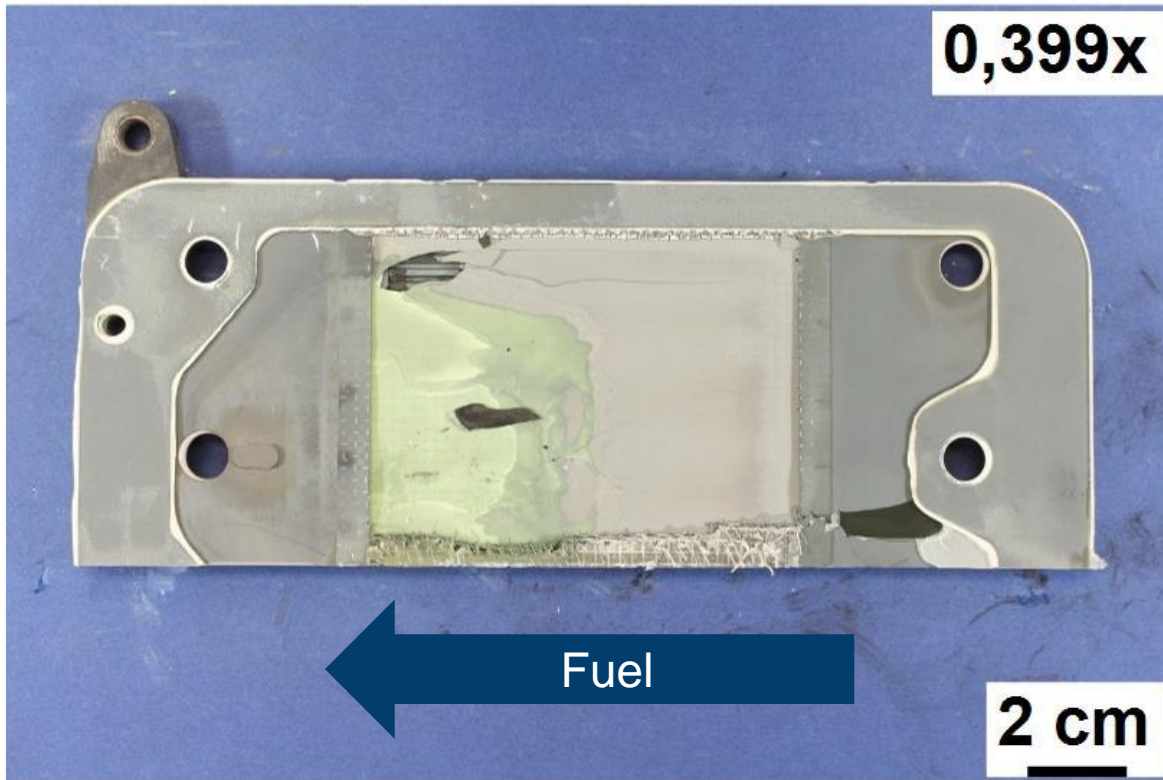
2-layer shortstack with PVD electrolyte



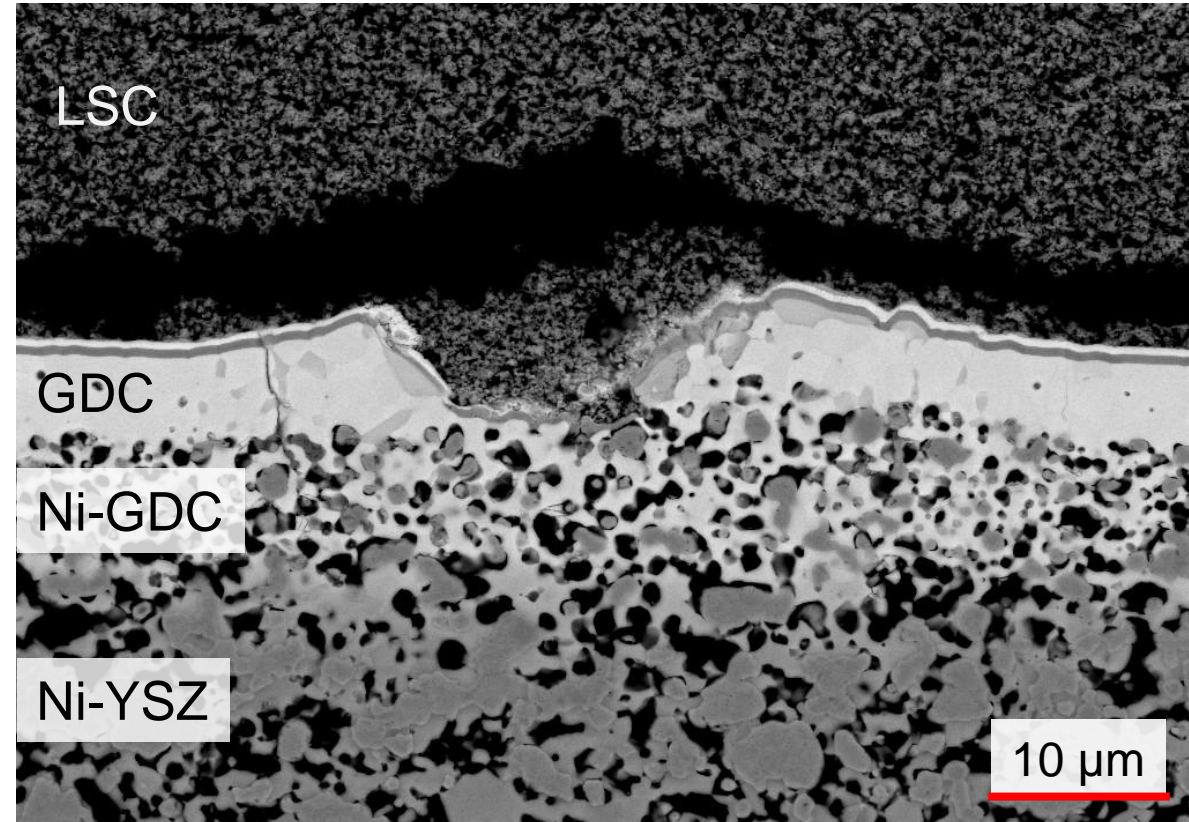
- 90% H₂O / 10% H₂ at inlet
- Excellent apparent performance
- Temperature increase below V_{TN}
- Negative differential resistance

2-layer shortstack with PVD electrolyte

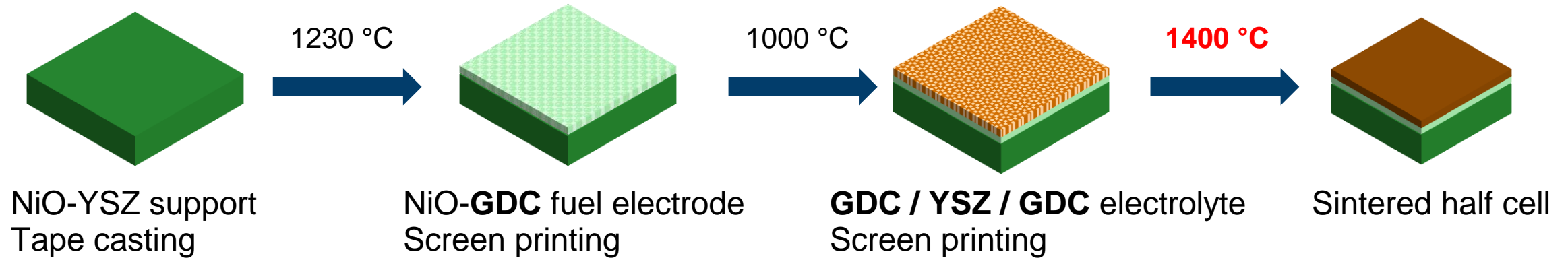
Image from stack opening – massive reoxidation



Air electrode touches GDC



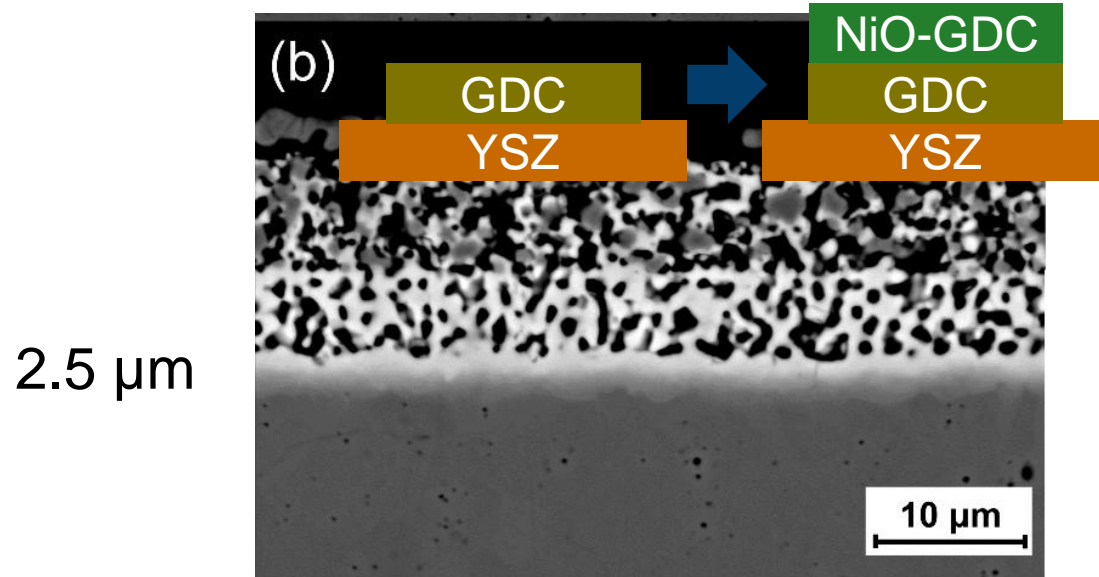
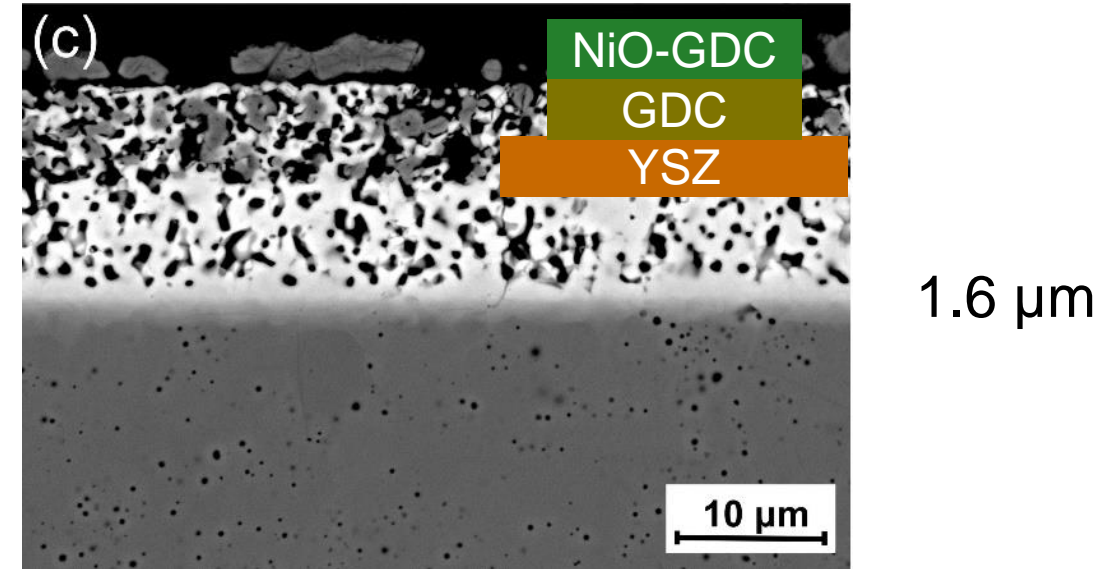
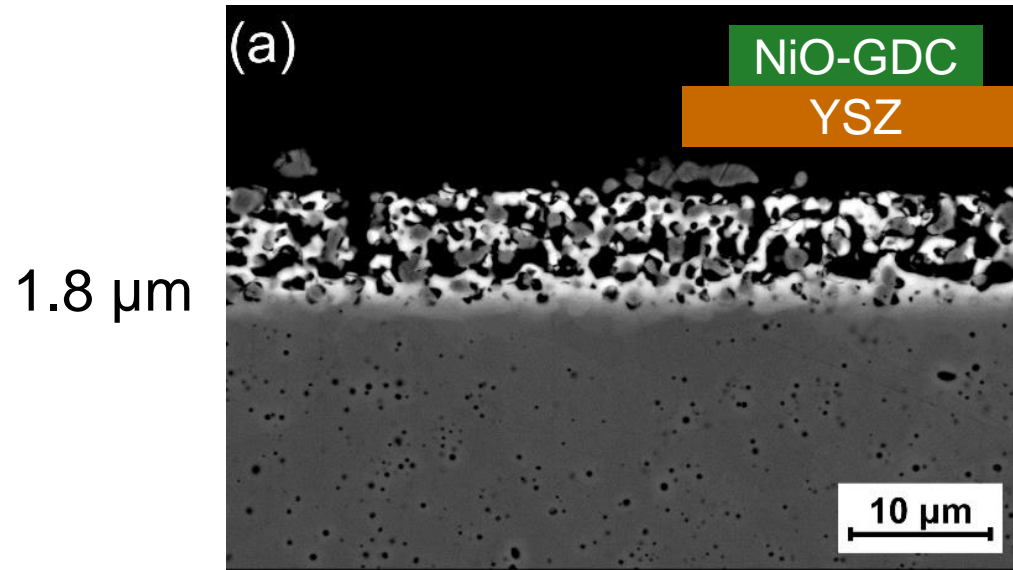
Screen-printing a three-layer electrolyte



- Fabrication of a three-layer electrolyte by co-sintering
- Layers must be matched in shrinkage during sintering and adapted to support
- Paste rheology must be controlled to achieve uniform and defect-free layers

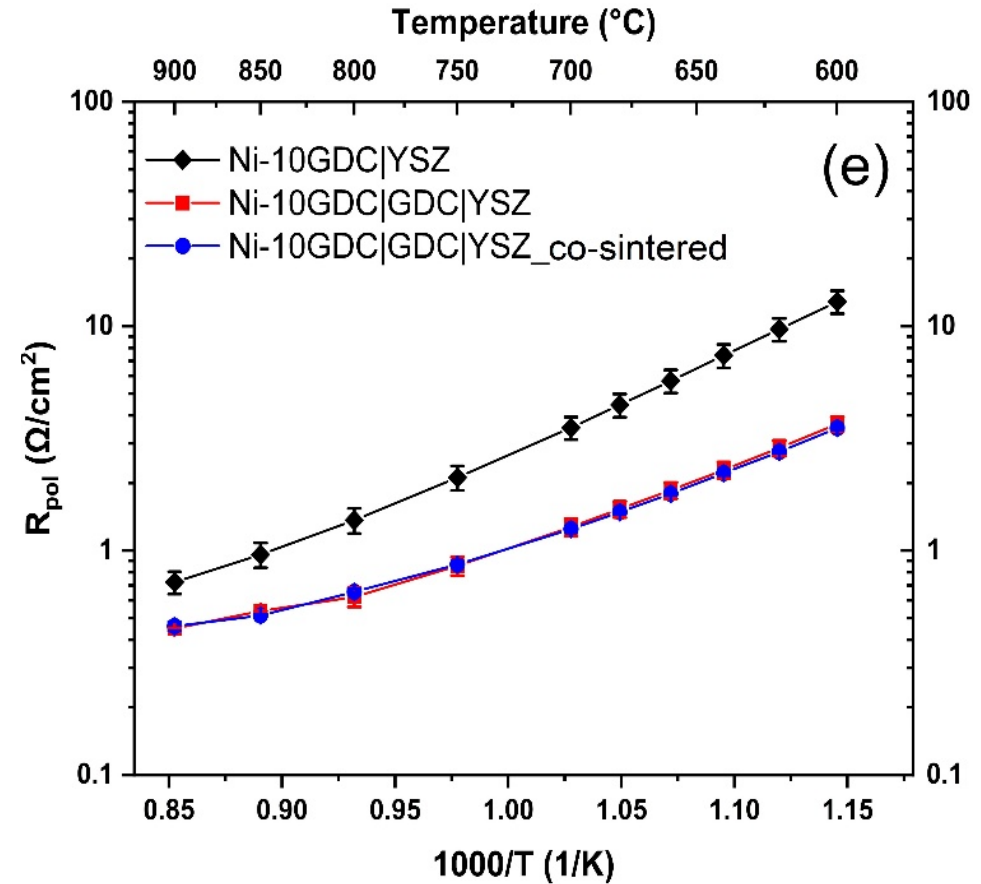
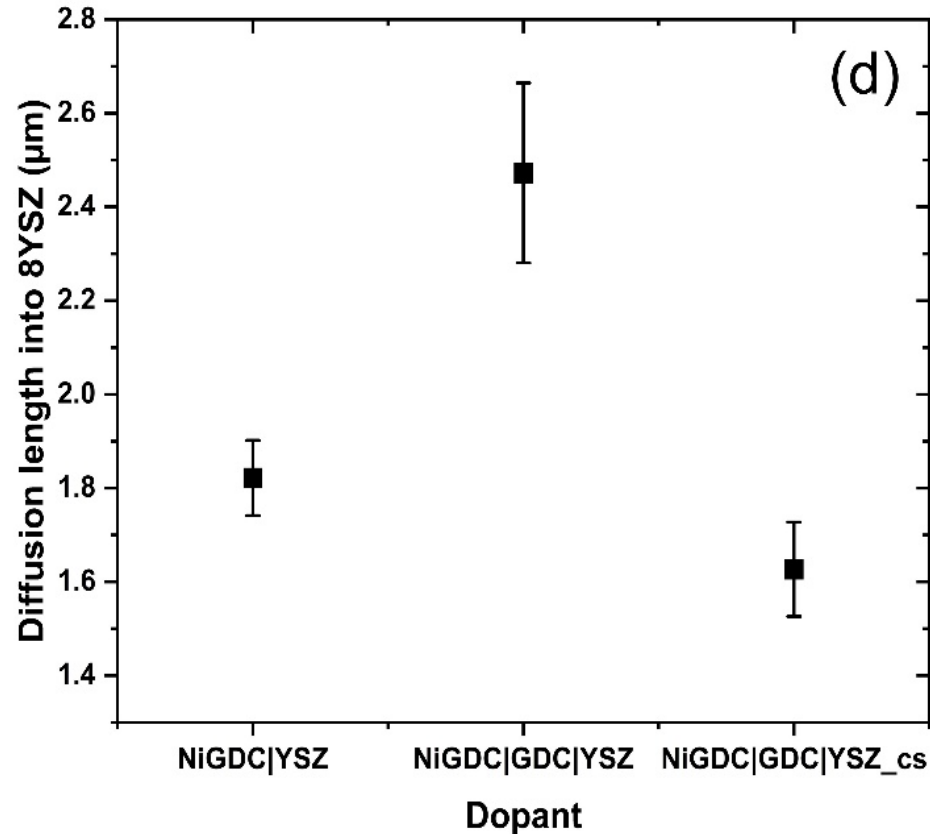
- But we know what we are doing – right?

Interdiffusion – sintering sequence



- Printed layers on dense YSZ ceramic
- Interdiffusion length scales with sintering T and duration
- Porosity at NiO-GDC; not at GDC-YSZ interfaces

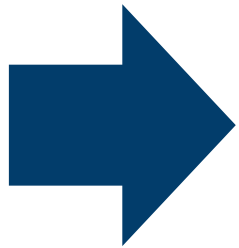
Interdiffusion – electrochemical impedance spectroscopy



- Increased R_{pol} for NiO-GDC sintered on YSZ
- Lower R_{pol} when GDC interlayer is used
- Not correlated to interdiffusion length, but to formation of porosity!

Co-firing strategy

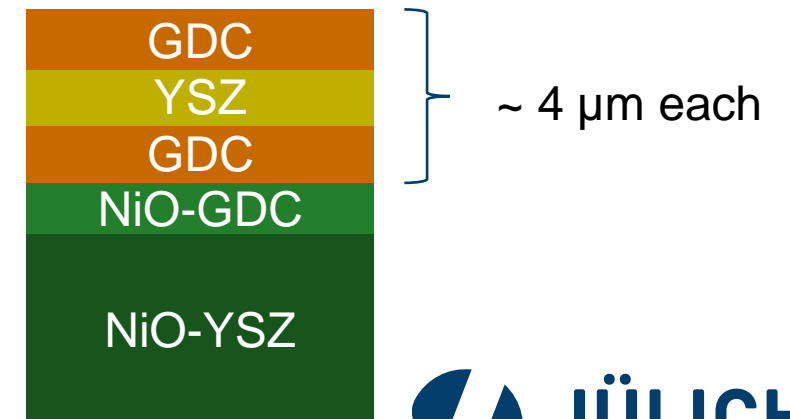
- Previous work shows:
 - Interdiffusion at NiO-GDC / YSZ leads to strongly increased cell resistance
 - Interdiffusion at GDC / YSZ interfaces only leads to small ohmic increases
 - Interdiffusion length during 1400 °C, 5h is smaller than 2 μm



Hypothesis:

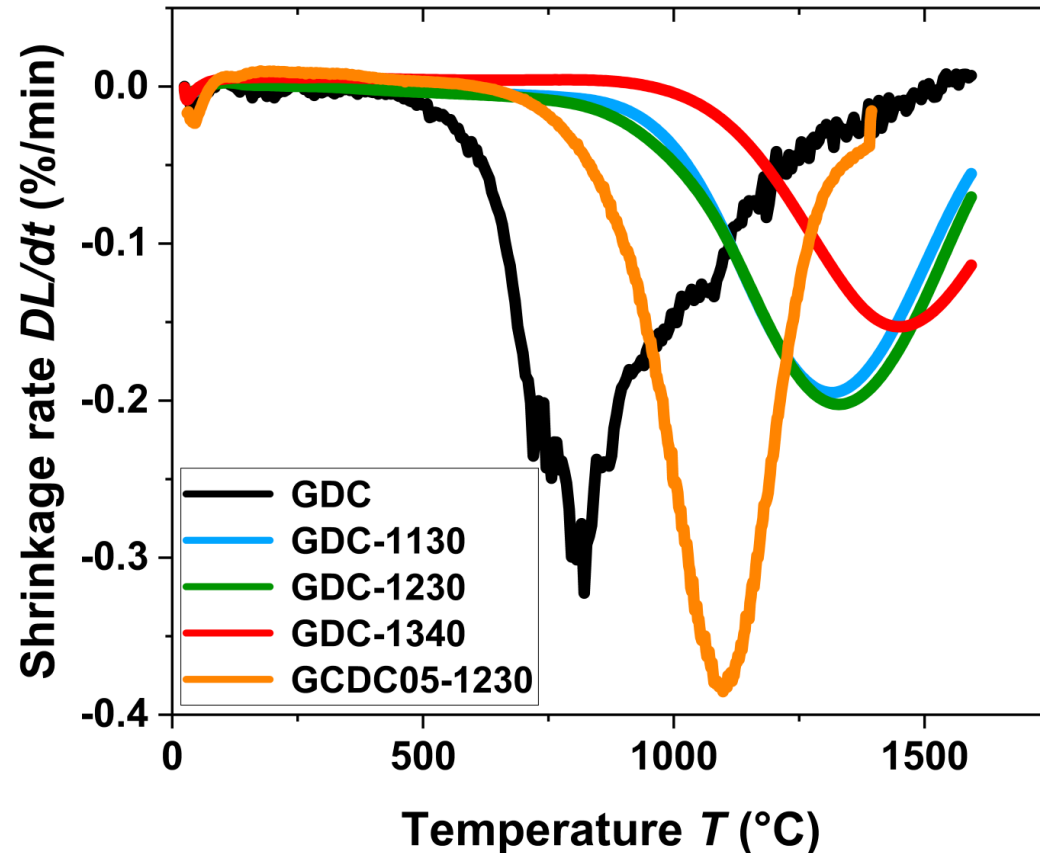
1) interdiffusion located inside the electrolyte should lead to acceptable performance losses!

2) a YSZ thickness of 4 μm should be sufficient.

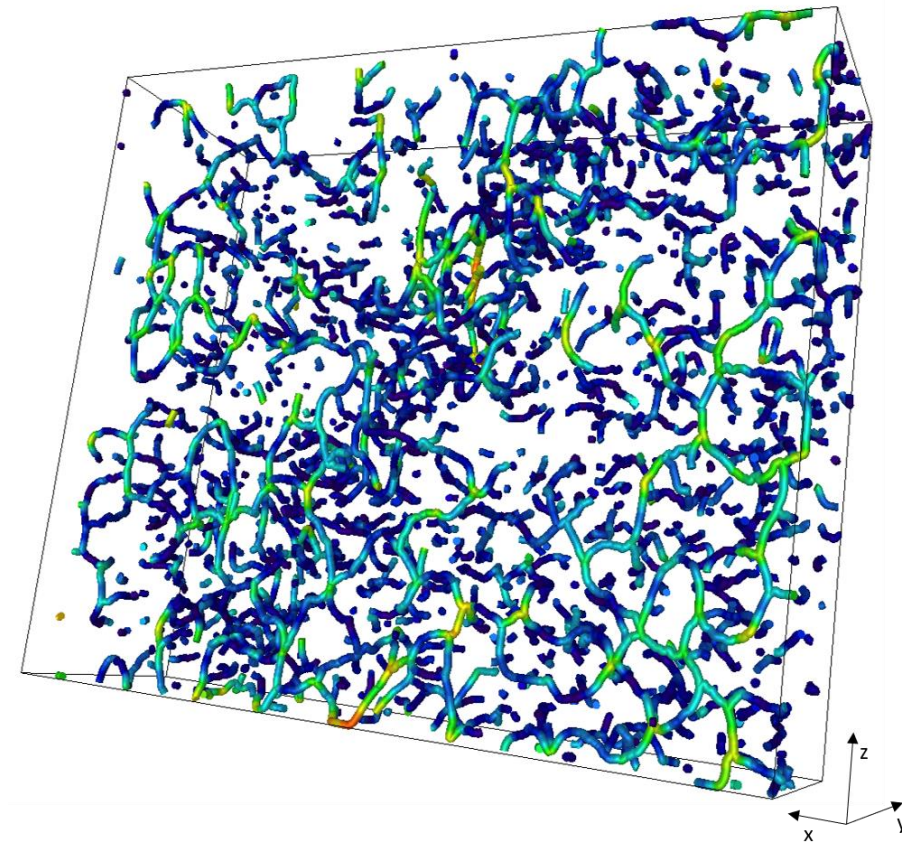
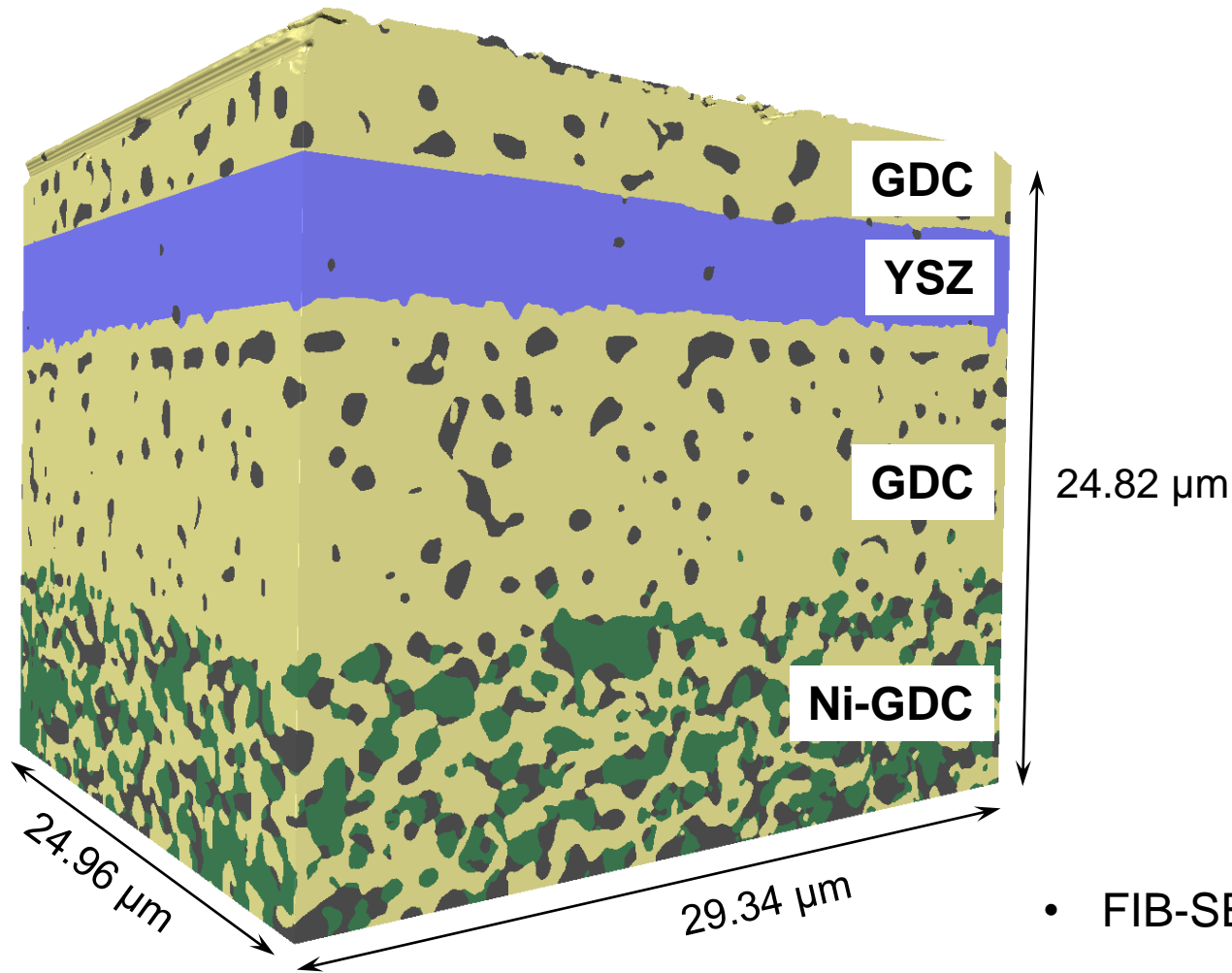


Screen-printing a three-layer electrolyte

- Adjusting shrinkage rate by powder calcination
- First results show complex sintering behavior – very dense YSZ, porous GDC

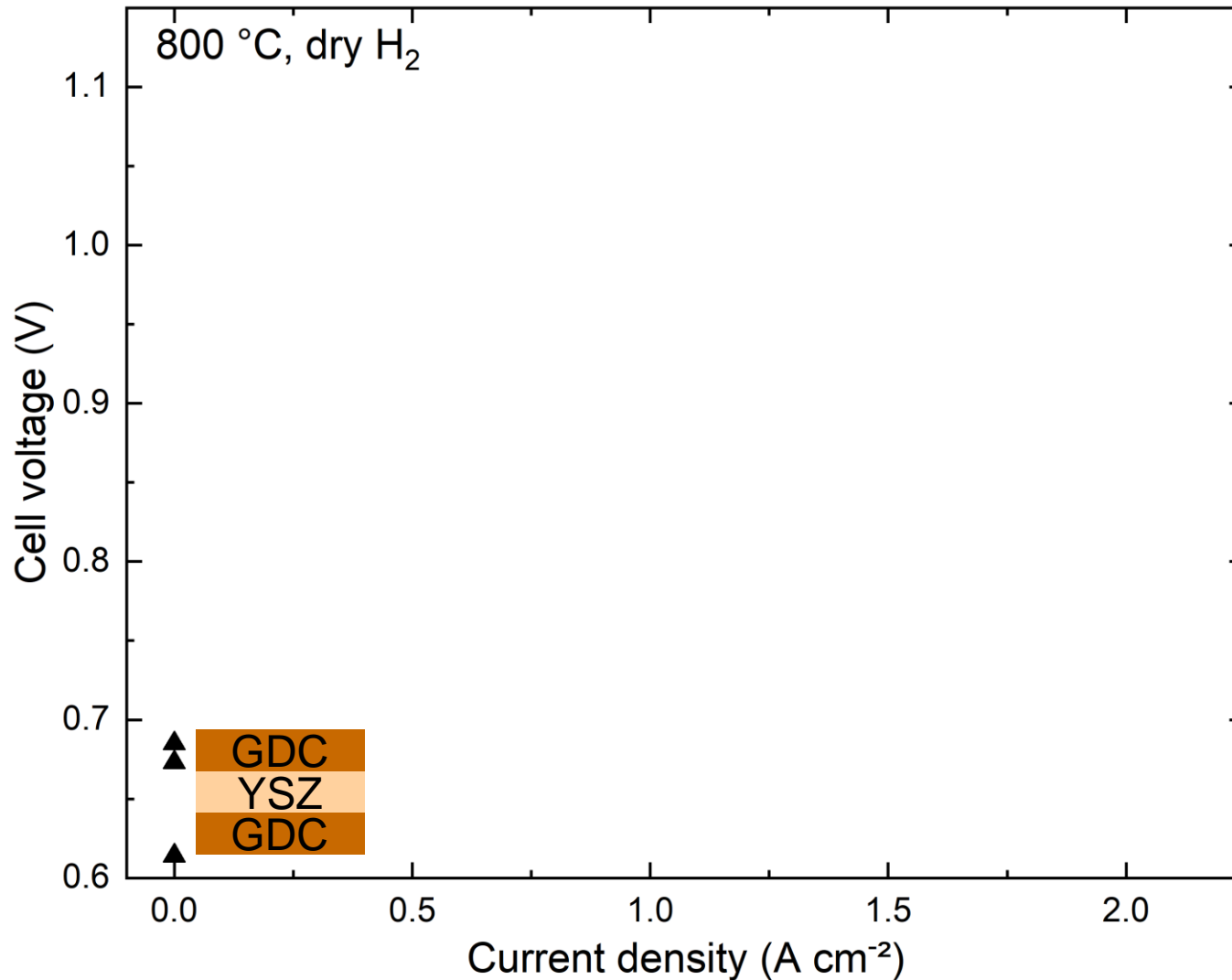


Ordered porosity after sintering



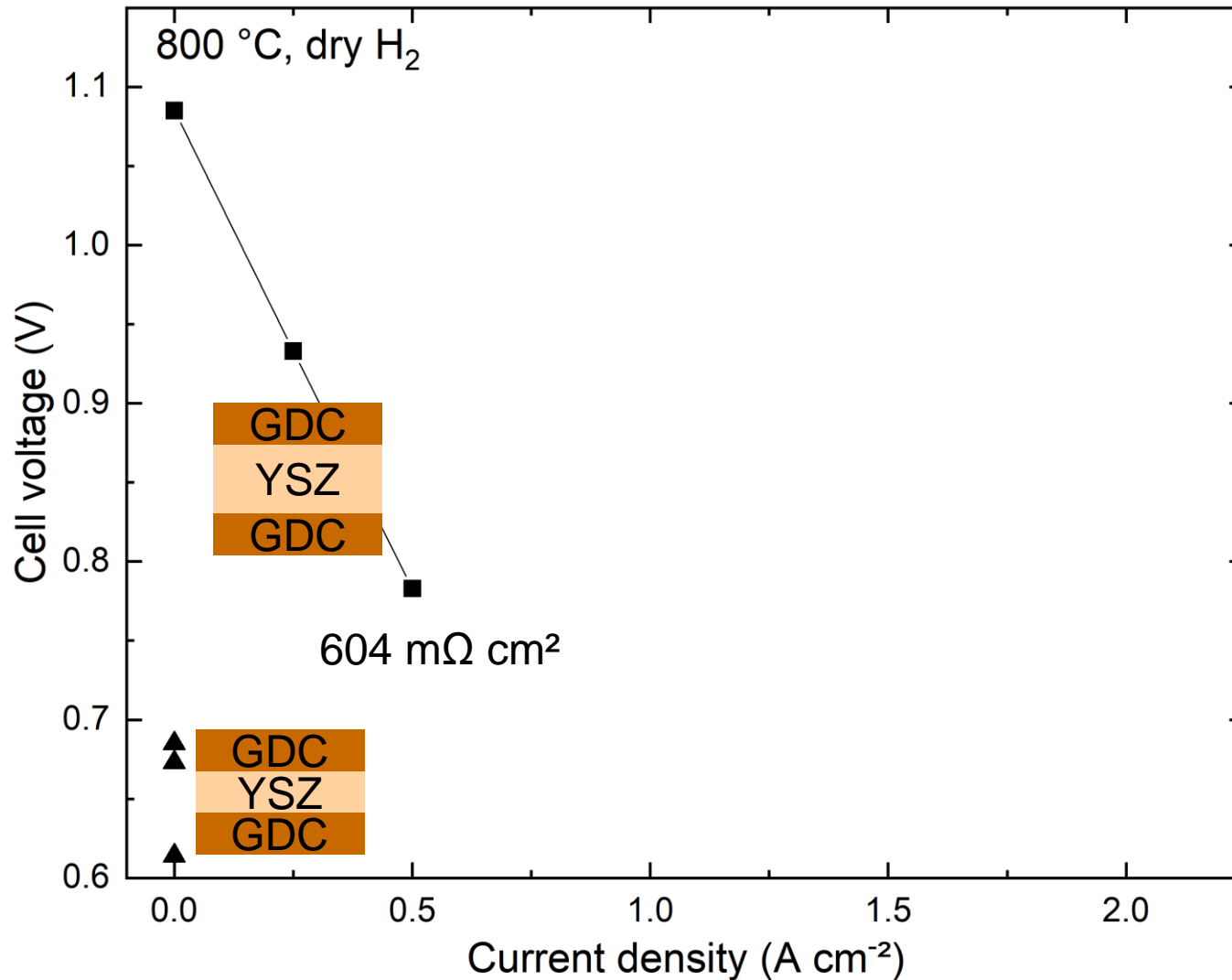
- FIB-SEM tomography confirms 2D connected pore network

Cell performance



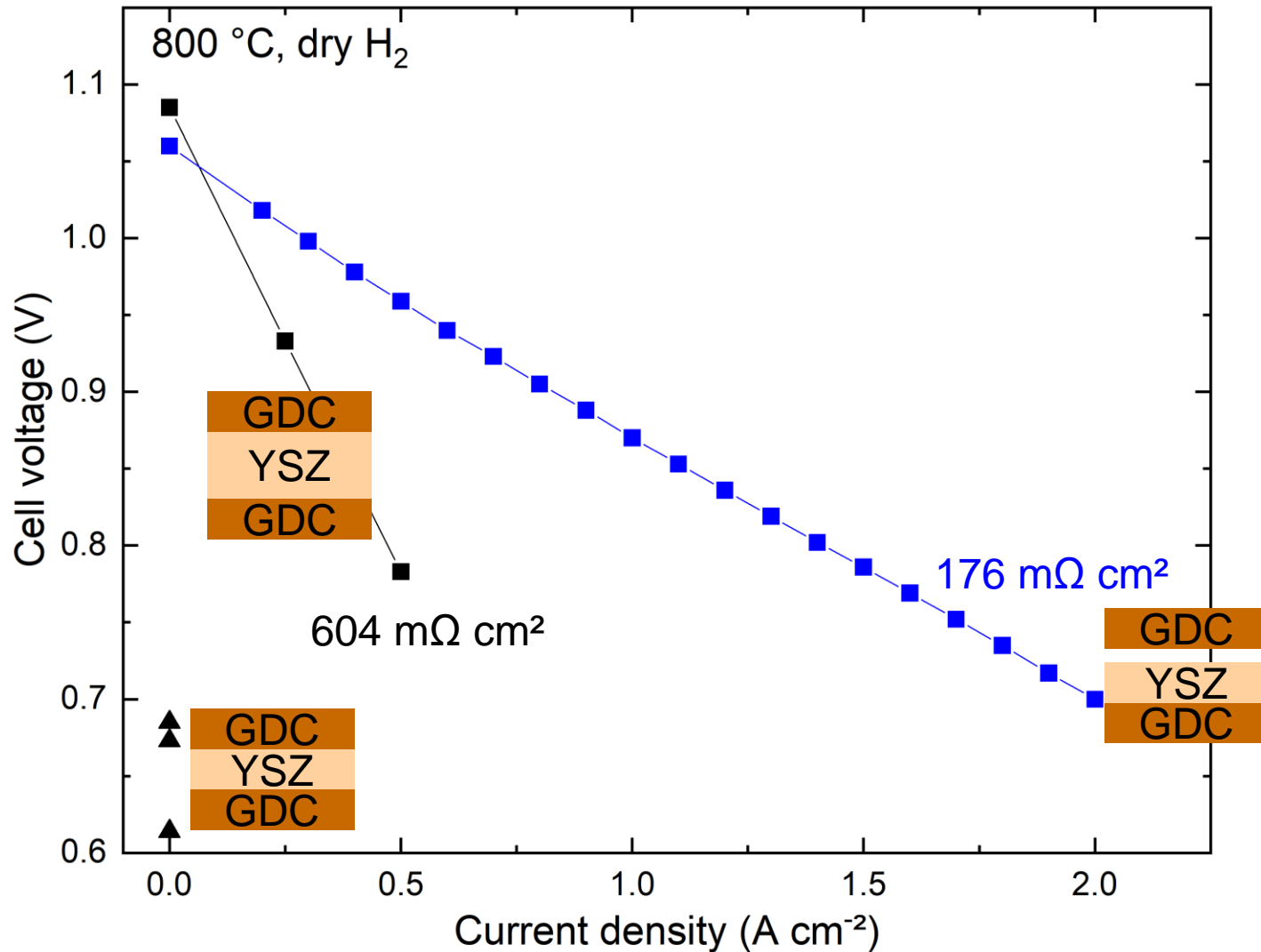
- Tested approx. 10 cells
- OCV values @ 800 °C, dry H₂ are between 0.6 and 0.7 V
- TEM investigations confirm that Ce diffuses through YSZ from both sides
 - Current leakage
- Best result for SP-cells when top GDC layer is sintered separately

Cell performance



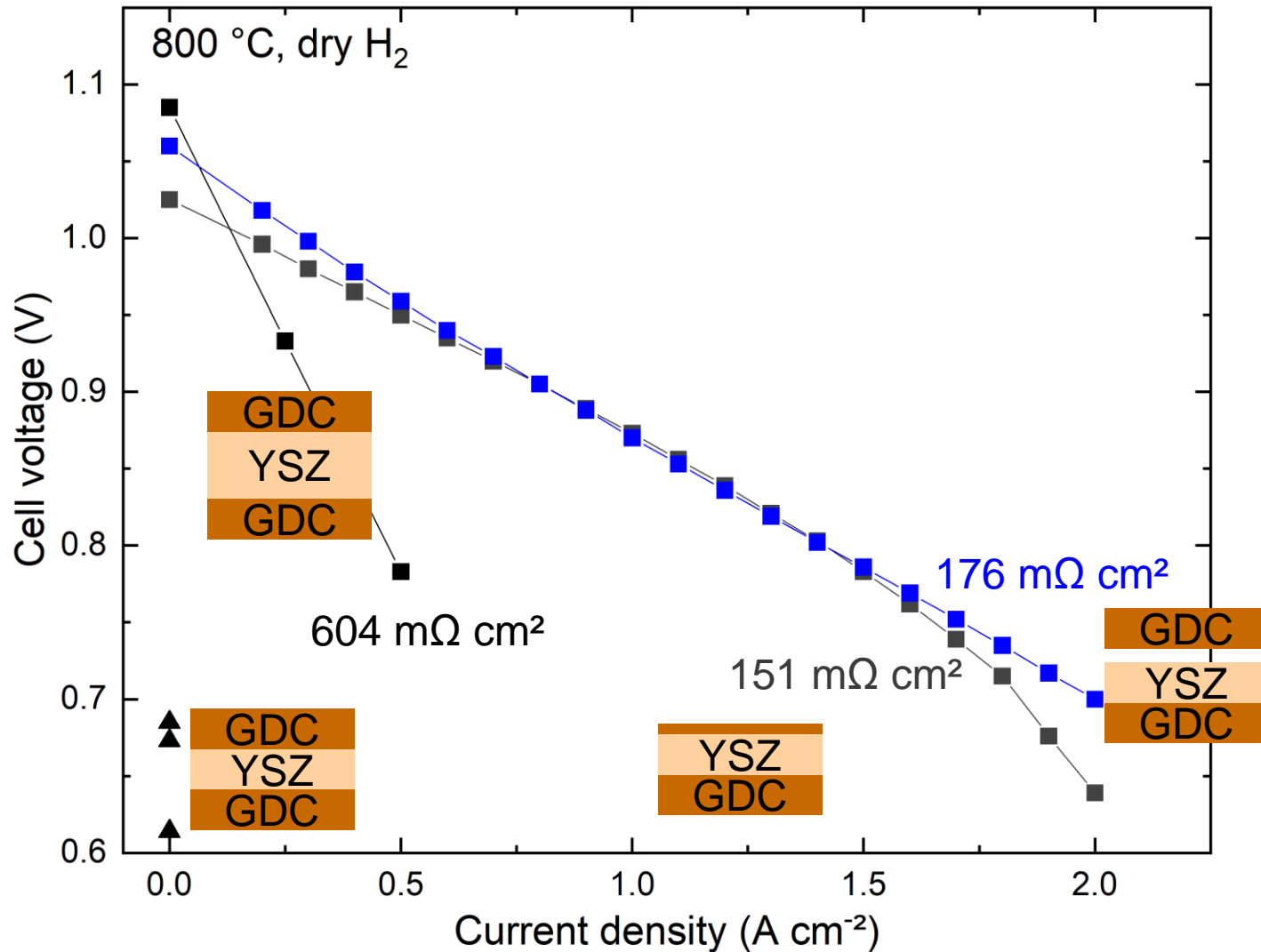
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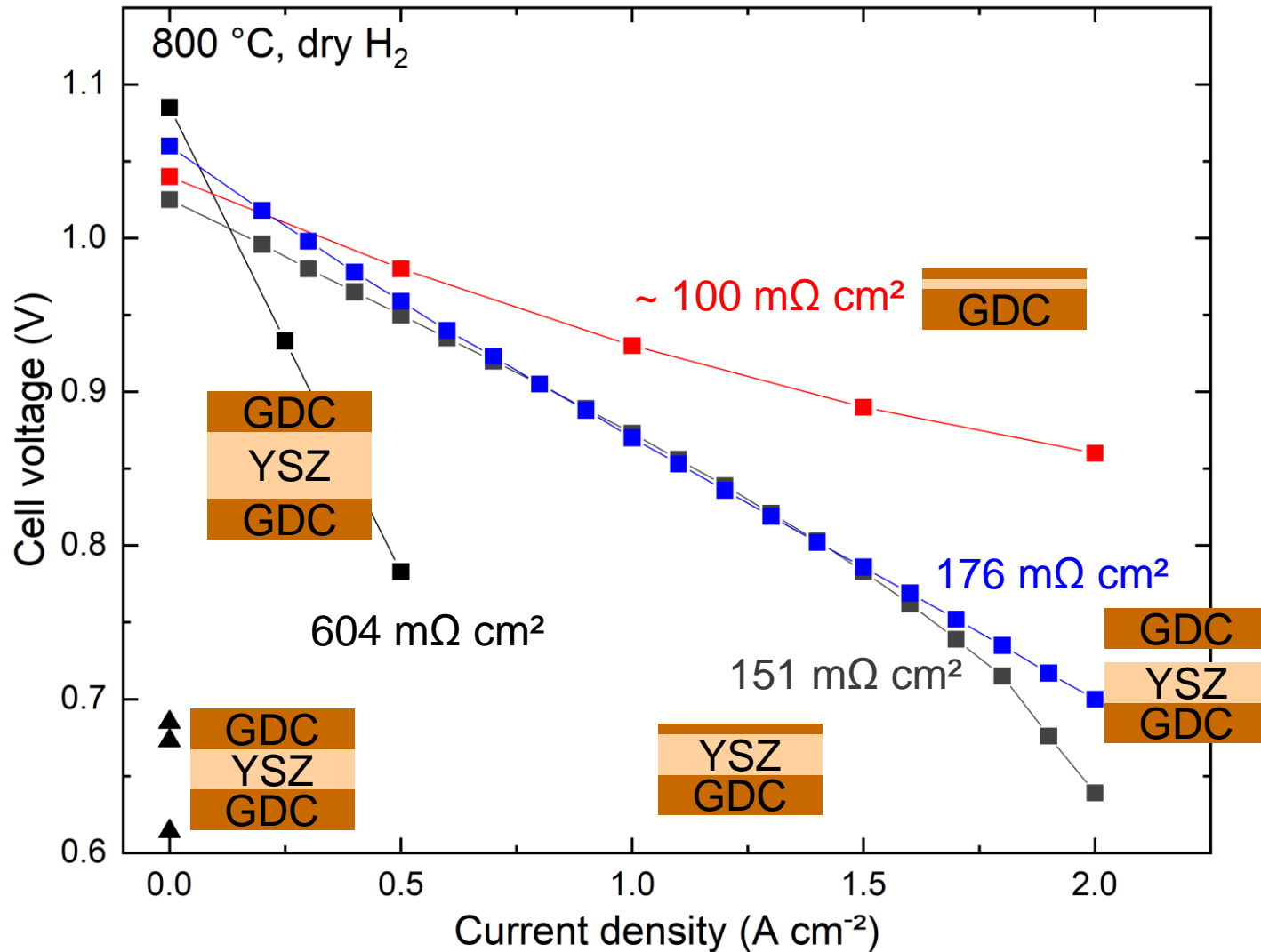
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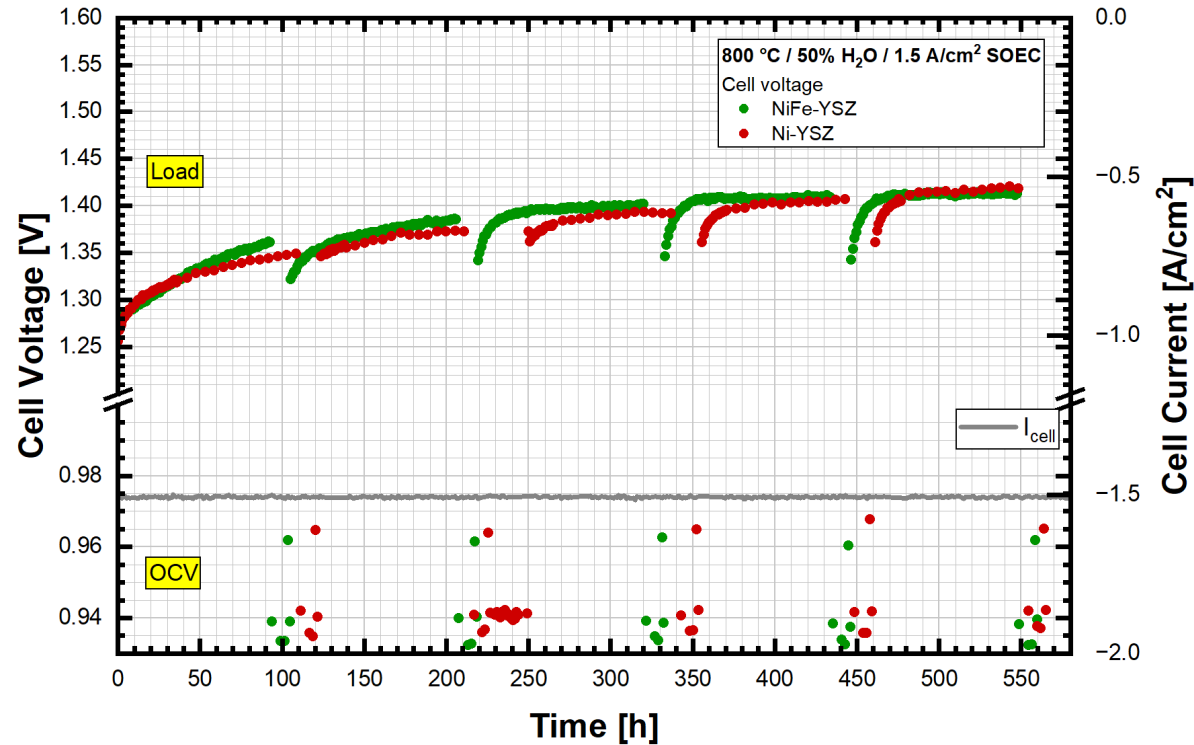
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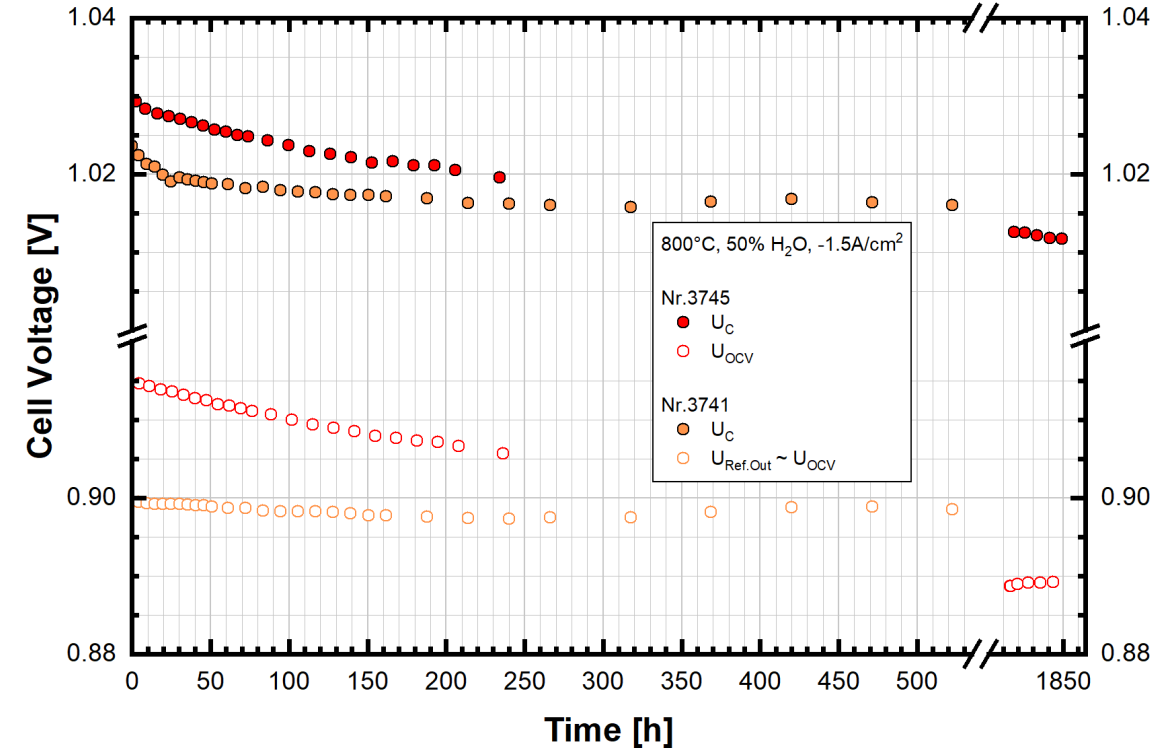
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Durability test – PVD cells

Baseline: Ni-YSZ & NiFe-YSZ

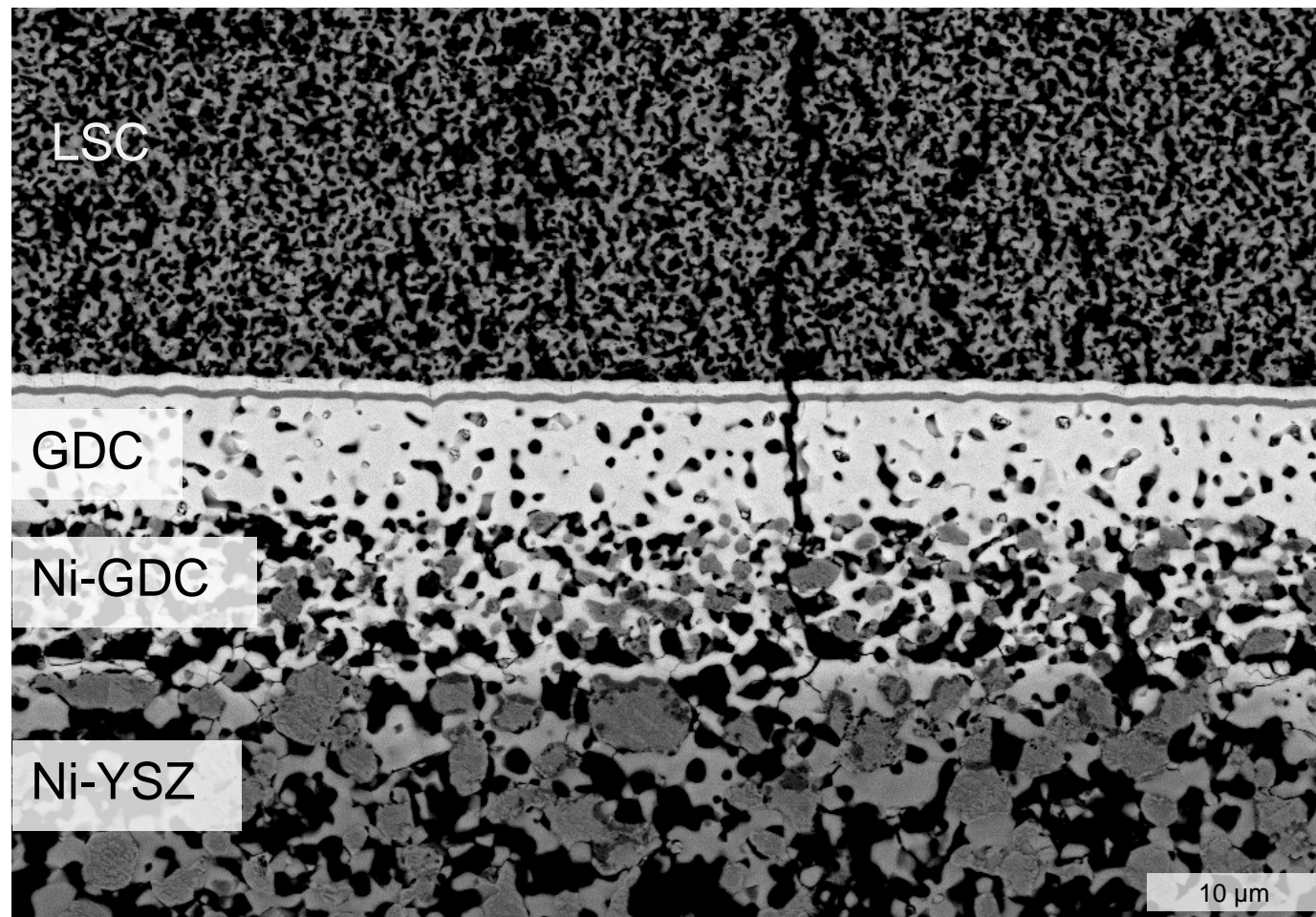
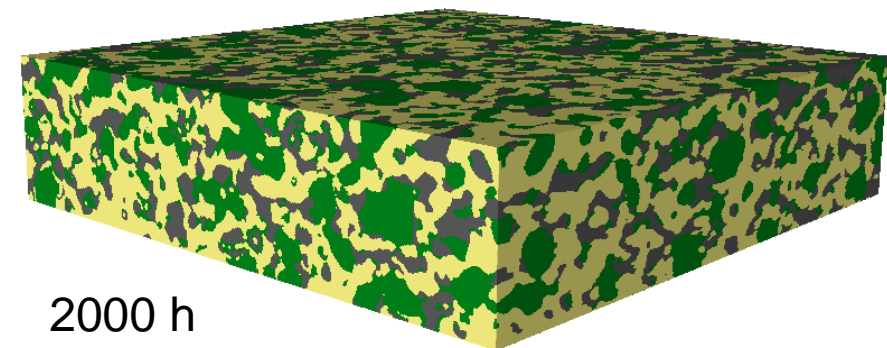
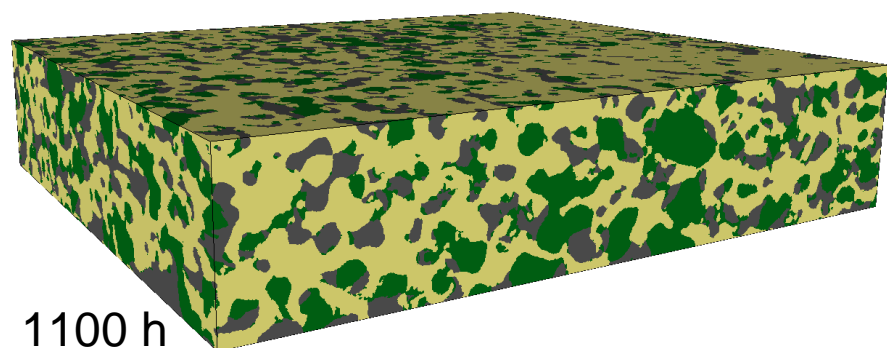
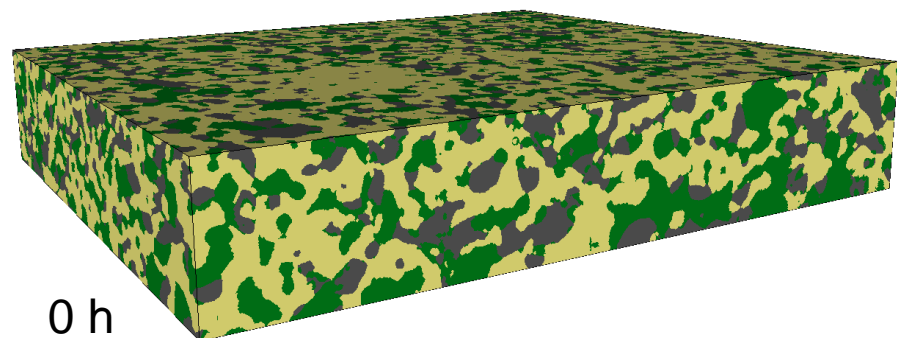


Ni-GDC with PVD electrolyte



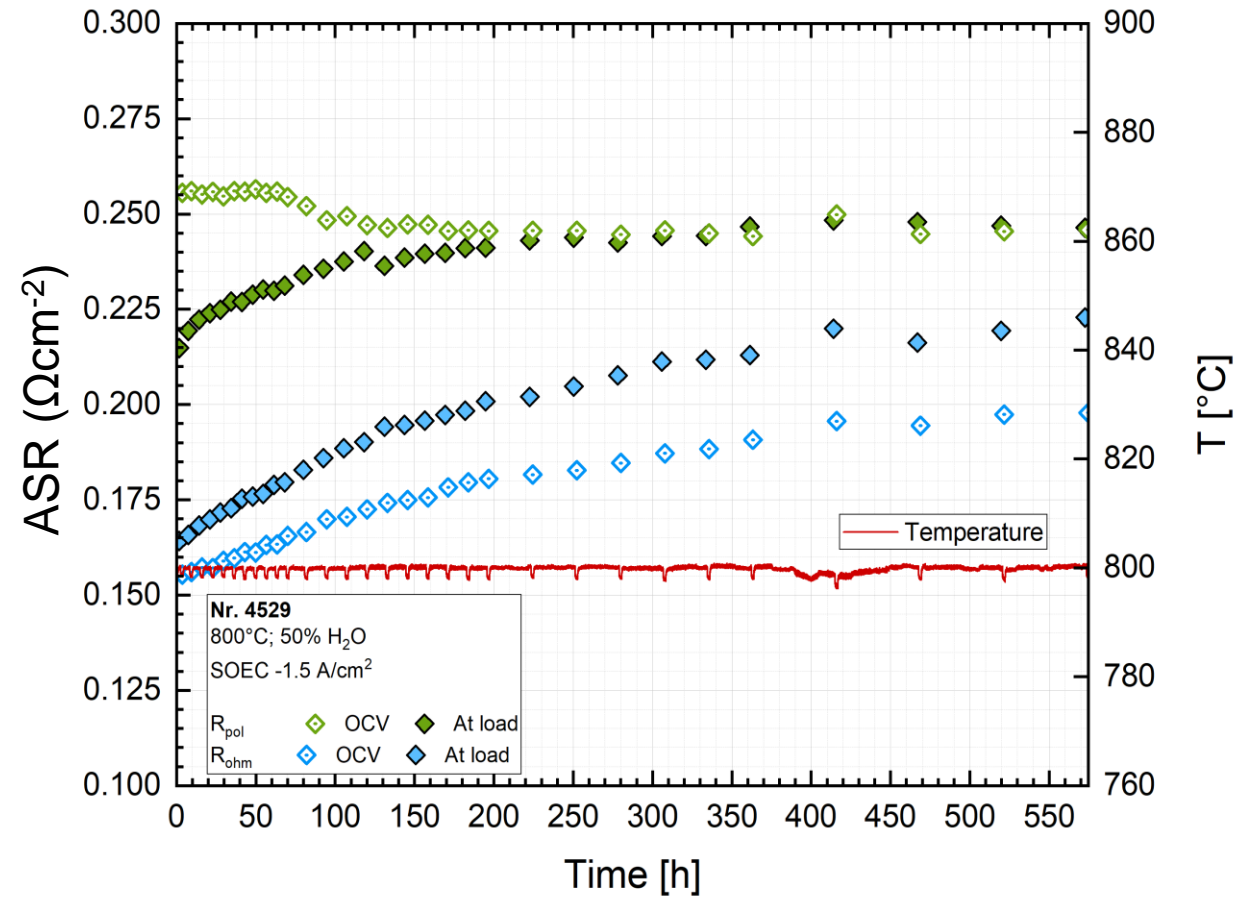
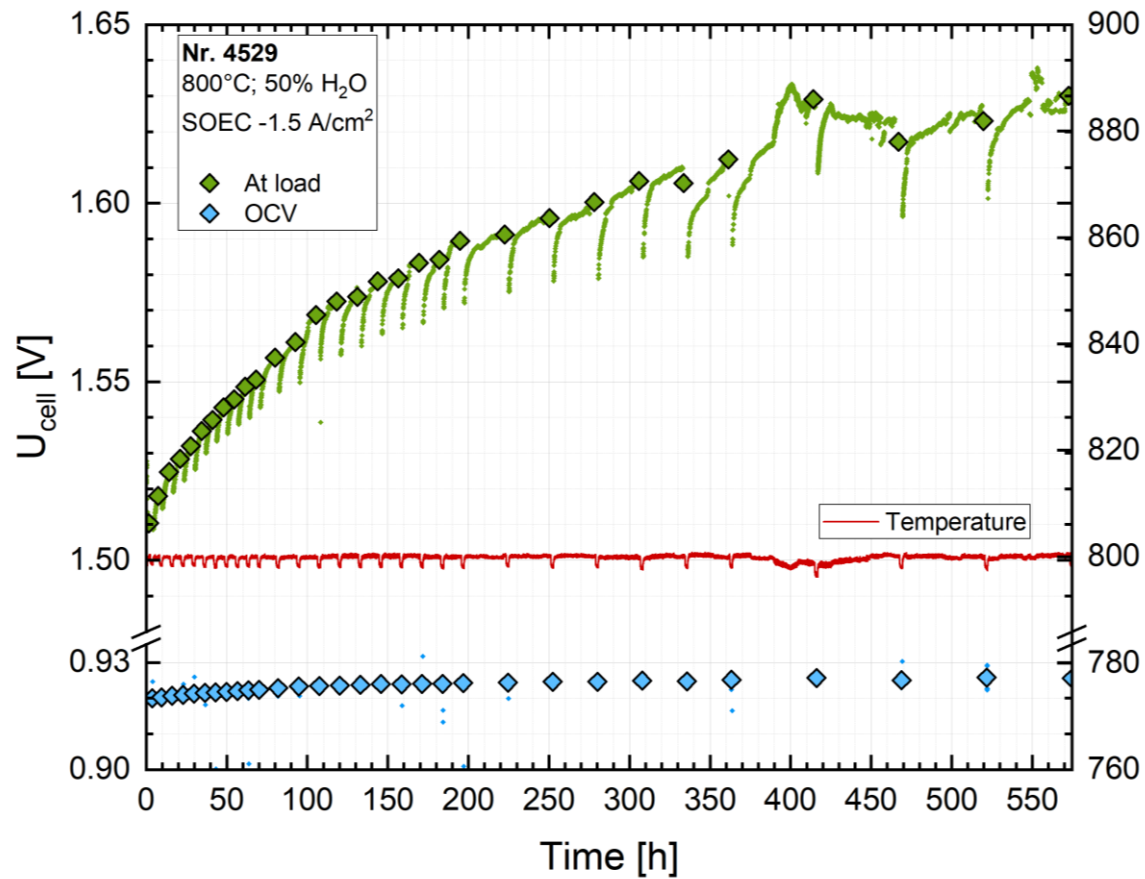
- 800 °C, 50:50 H₂/H₂O, 1.5 Acm⁻²: Ni(Fe)-YSZ shows strong degradation
- Ni-GDC very stable over 2000 h, slight decreasing in (open circuit) voltage
- But: Faradaic efficiency unknown!

Post-test PVD cells: FIB-SEM reconstructions



26

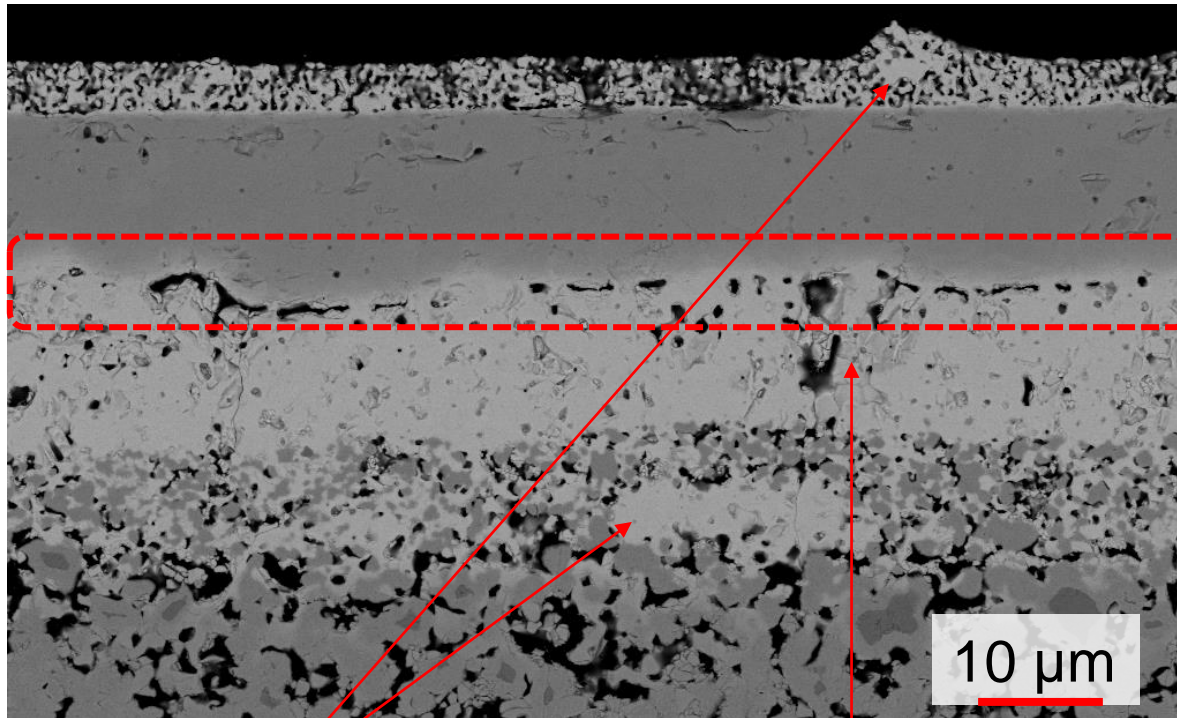
Durability test – SP cell



- 800 °C, 50:50 H₂/H₂O, 1.5 Acm⁻²: Substantial degradation; mostly ohmic ASR

SP- cell: Microstructure

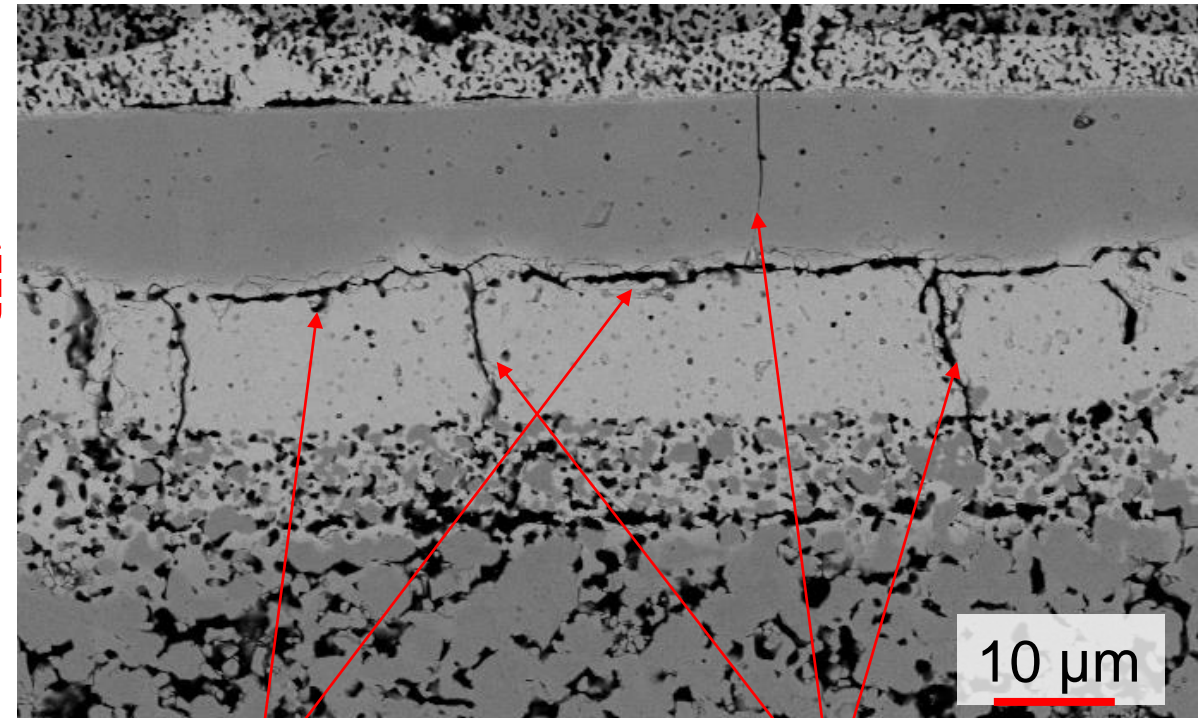
Reduction only



Agglomerates

Interface porosity

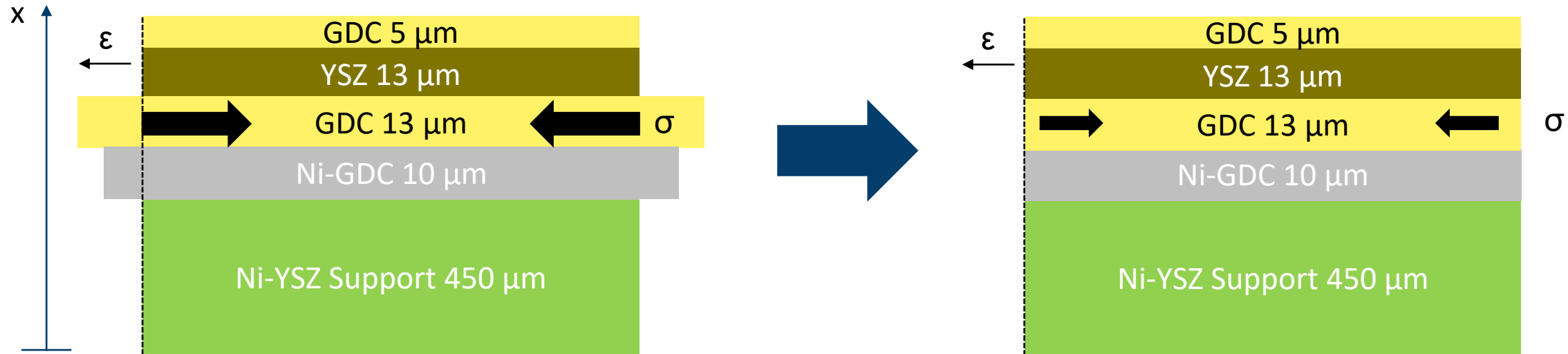
After testing



Delamination

Cracks

Microstructure degradation: hypothesis



Electro-chemical expansion of GDC under cathodic polarization:
GDC is constrained \rightarrow compressive stress

Compressive stress relaxes through creep

Removing polarization induces stress inversion
 \rightarrow tensile stress leads to cracking

Summary

- Ni-GDC was successfully integrated into an ASC with YSZ blocking layer
- PVD-YSZ requires near-perfect GDC microstructure
 - Pores in GDC result in high leakage currents & gas leakage
 - Performance & durability with improved GDC is exceptional, BUT Faradaic efficiency not known
- Co-firing on GDC-YSZ-GDC requires ~ 15 μm YSZ to prevent current leakage
 - High resistance & degradation due to very limited contact between GDC + YSZ
 - Much lower sintering temperature required to solve this problem

Acknowledgements



Alex Schwiars



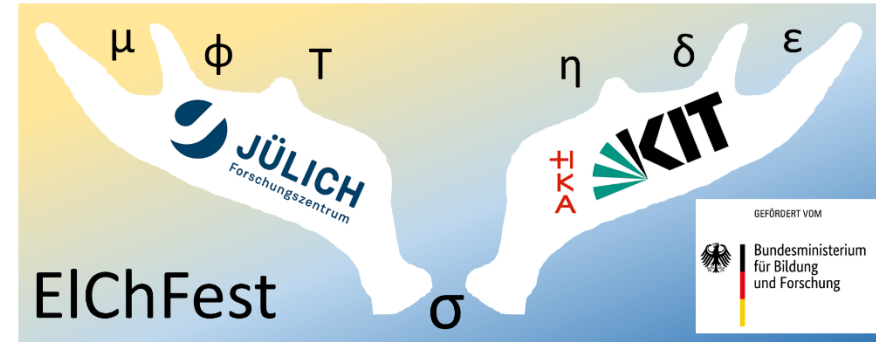
Denise Ramler



Luzie Wehner



Iurii Kogut



Grant no: 03SF0641A

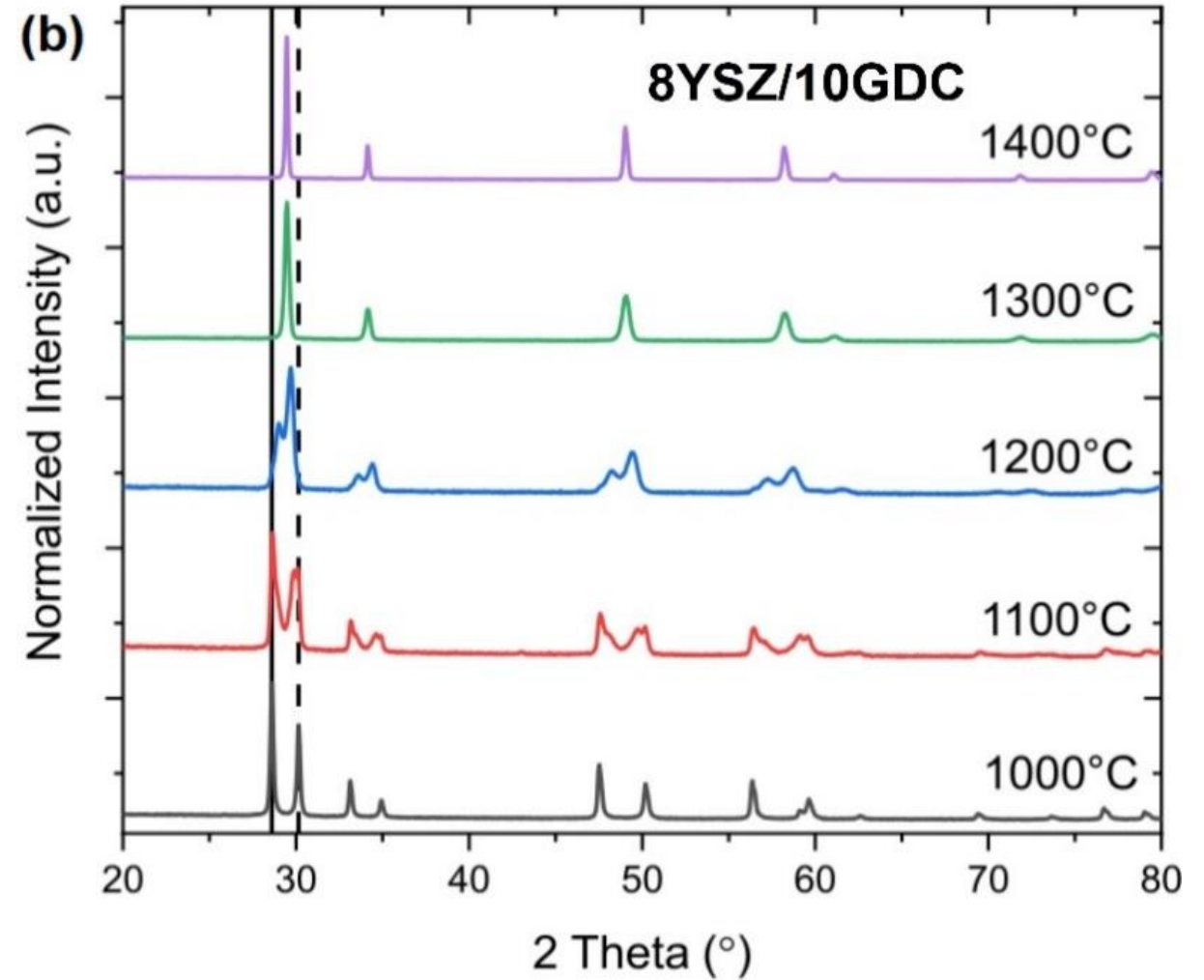
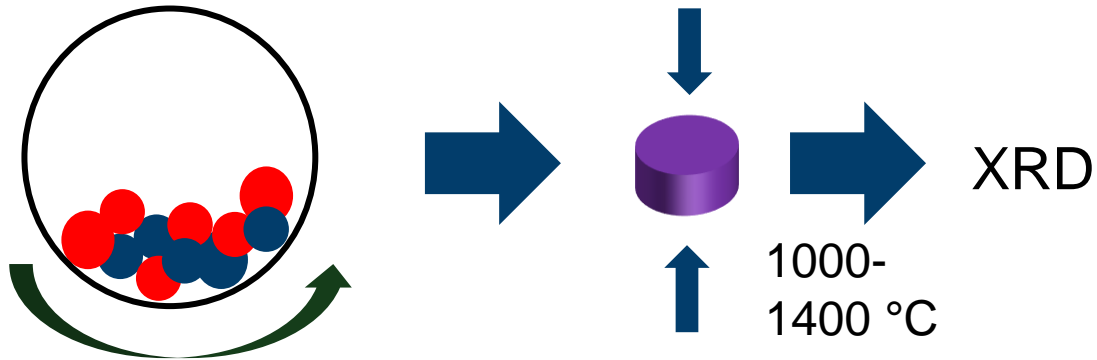


Grant no: 275388933

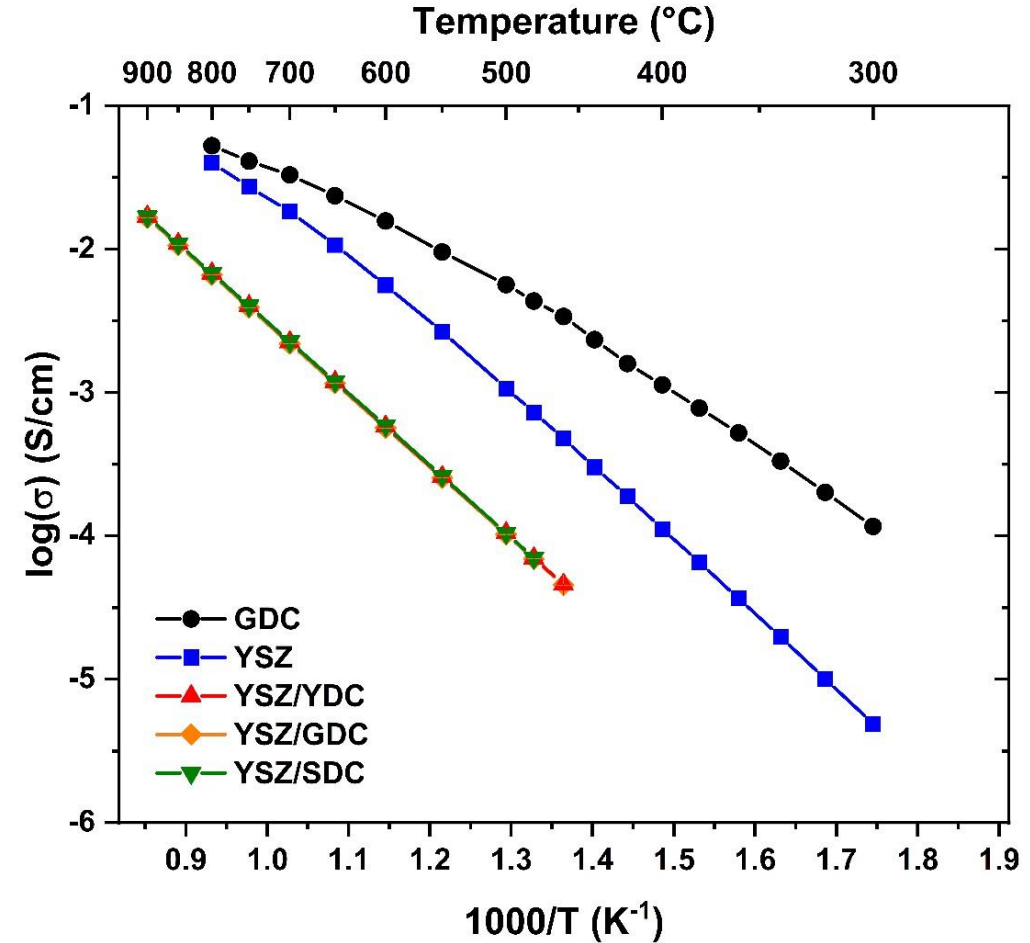
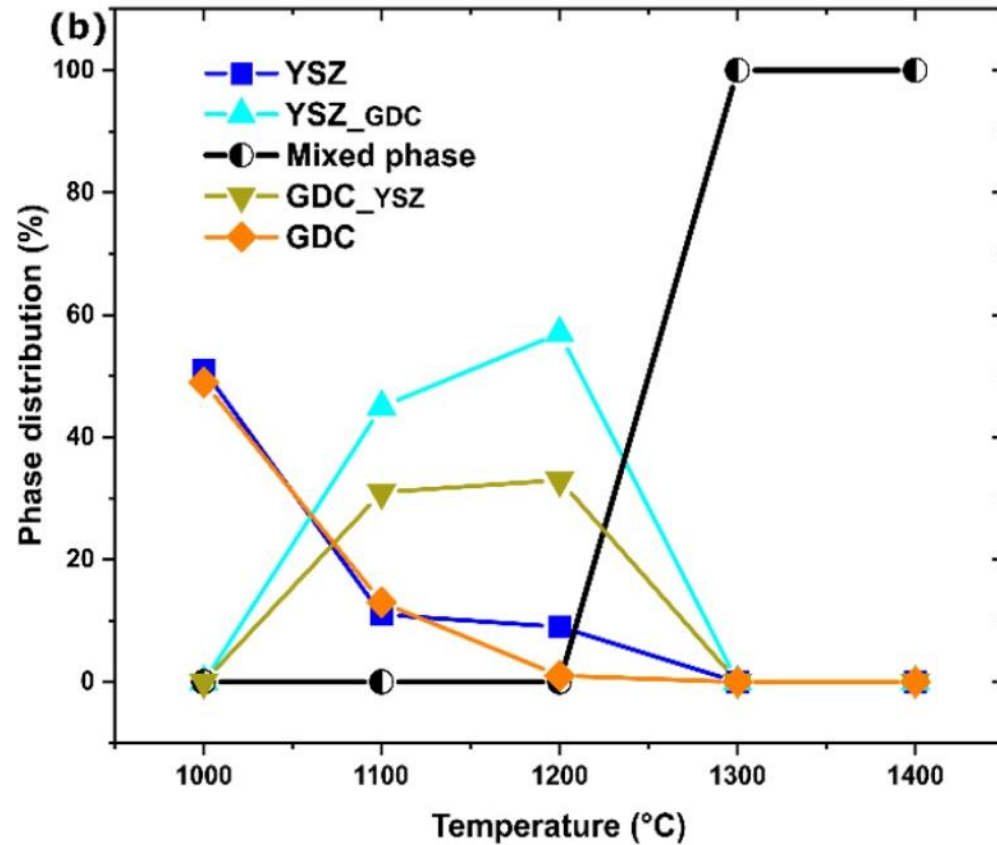
Thank you for your attention!

Interdiffusion experiments

- Started these experiments in 2018
- Sinter well mixed powders
 - GDC: $d_{50} = 0.1 \mu\text{m}$
 - YSZ: $d_{50} = 0.48 \mu\text{m}$
- Crush pellets + phase analysis via XRD

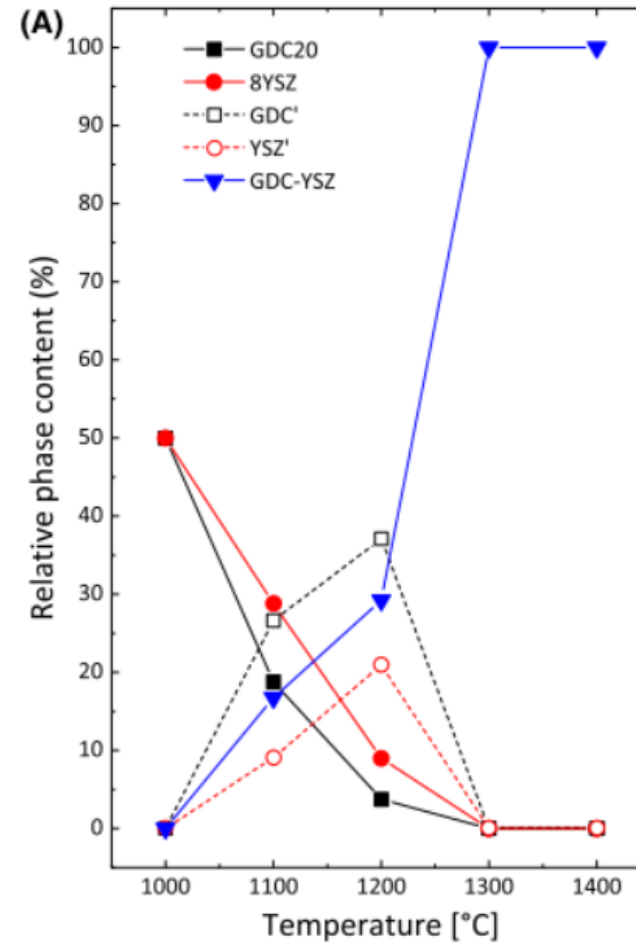
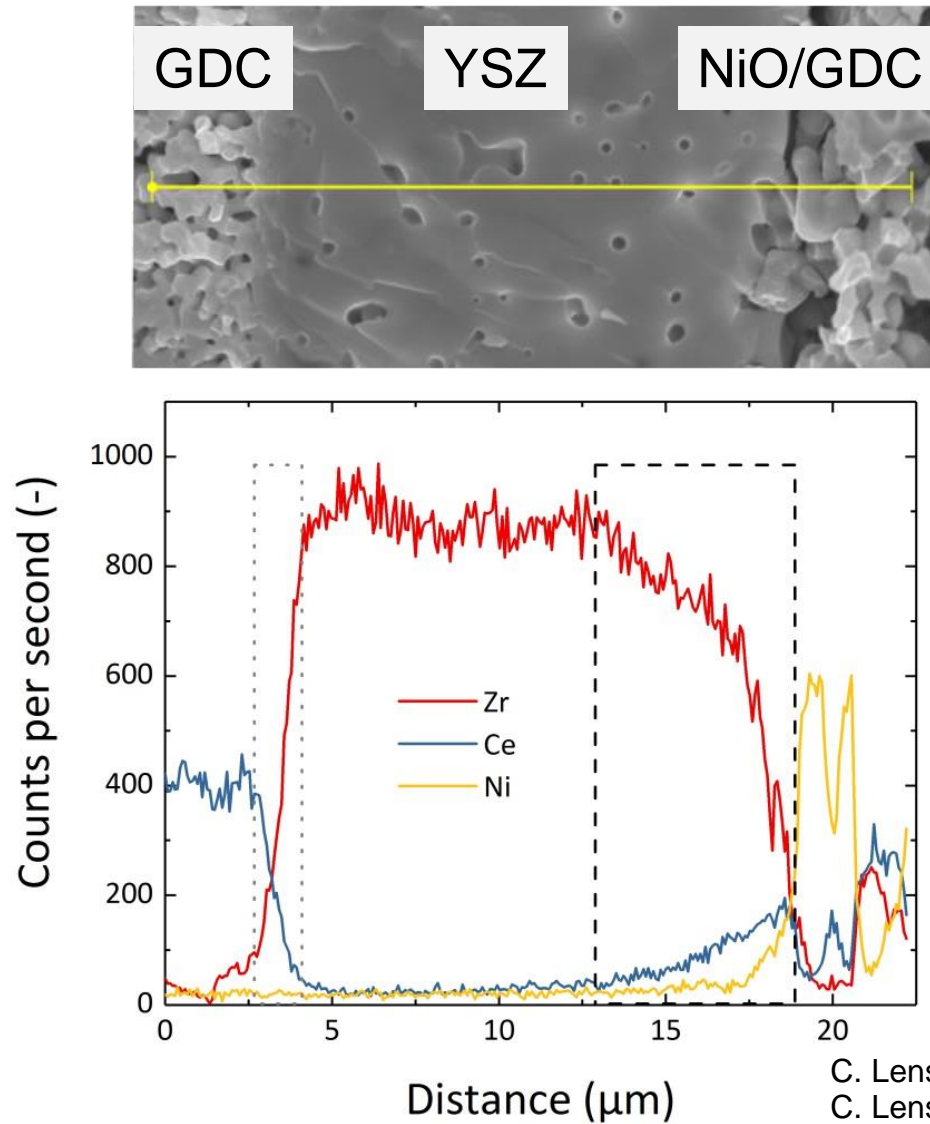


Interdiffusion – phase composition



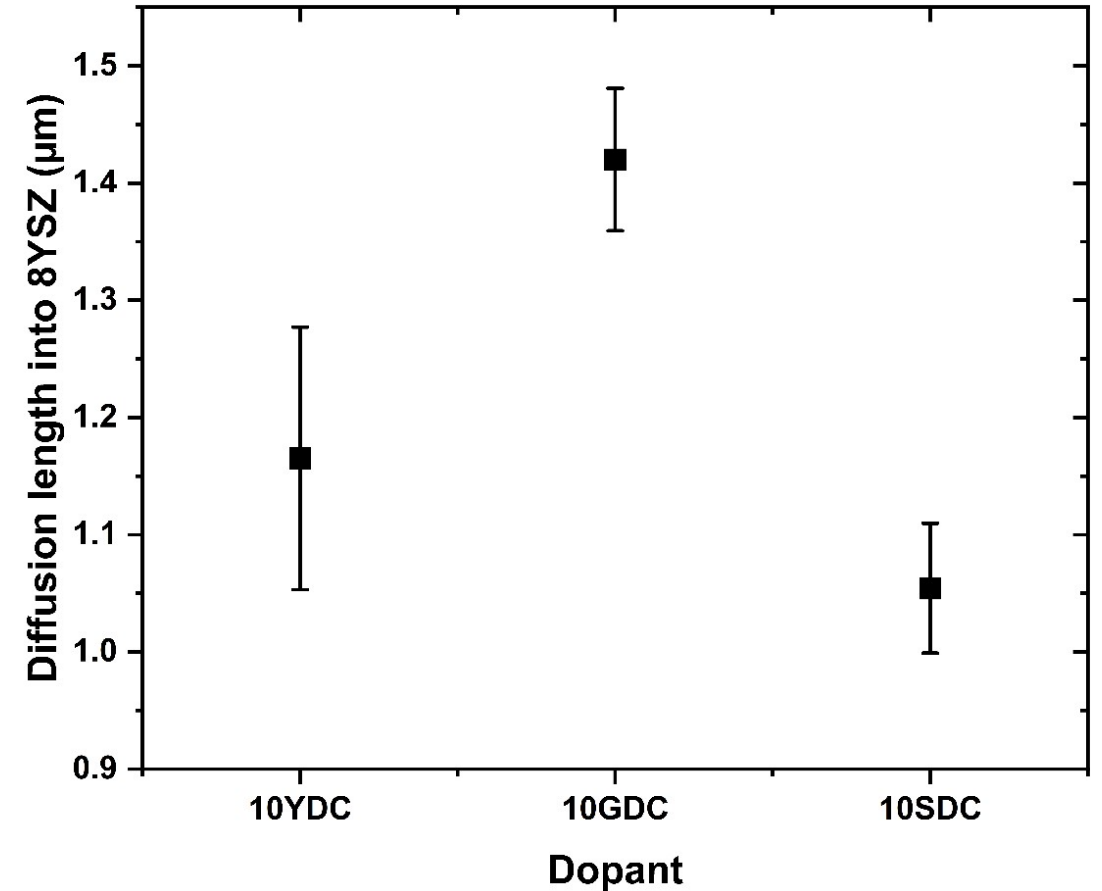
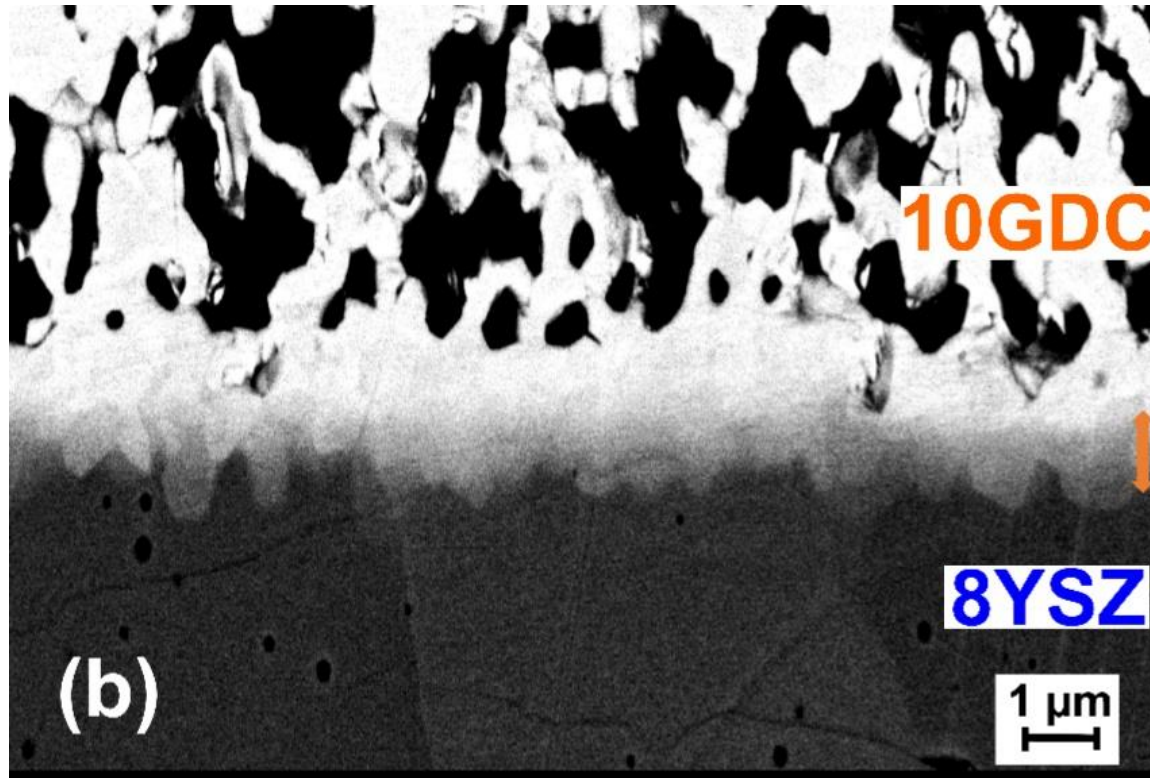
- Complete mixing after 1400 °C, 5h
- Decreased conductivity as reported in literature

Interdiffusion - the role of NiO



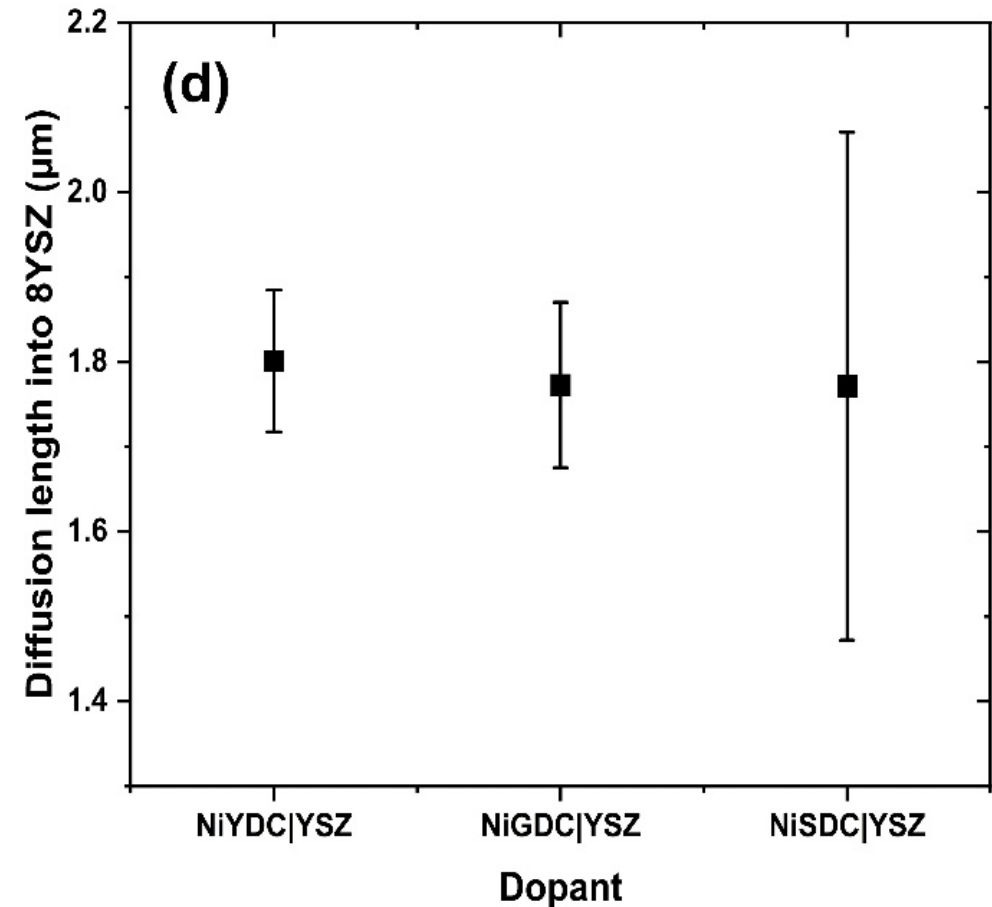
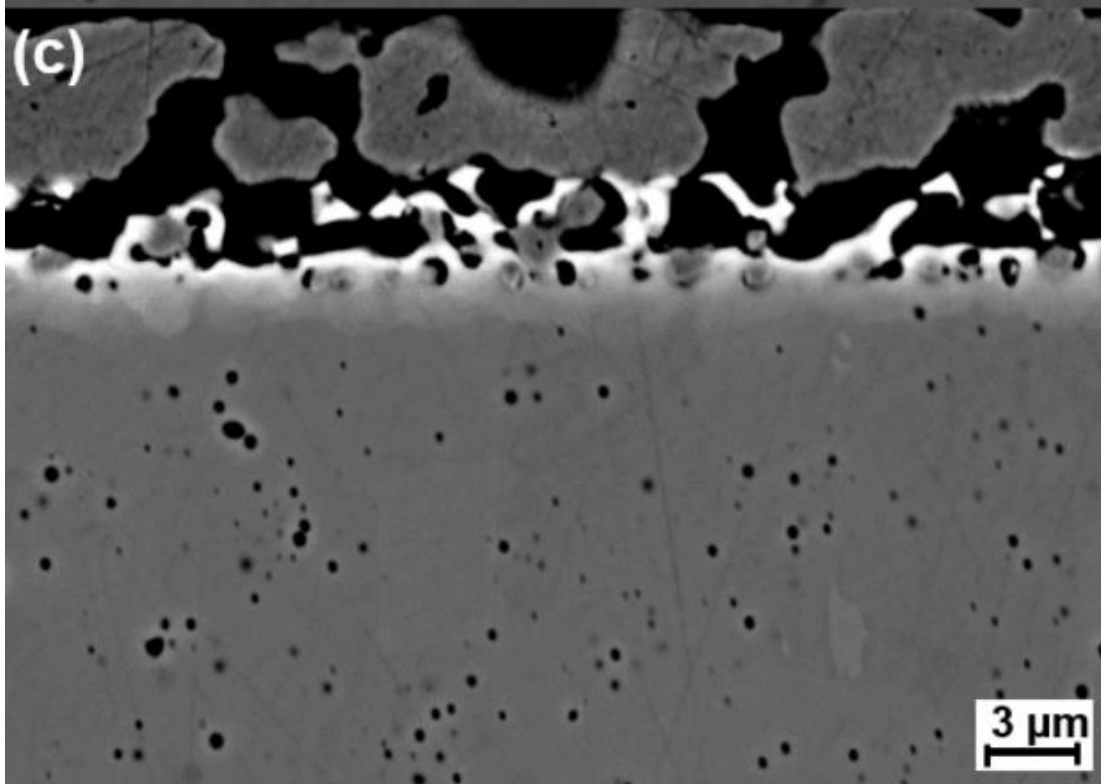
C. Lenser et al., Journal of the American Ceramic Society, 101 (2017) 739-748.
 C. Lenser, N.H. Menzler, Solid State Ionics, 334 (2019) 70-81.

Interdiffusion length – GDC on YSZ



- Printed layers on densified ceramic
- SEM analysis using Z-contrast show interdiffusion length 1 – 1.5 μm

Interdiffusion length – NiO-GDC on YSZ



- Printed layers on densified ceramic
- SEM analysis using Z-contrast show higher interdiffusion length ~ 1.8 μm
- Notable porosity is formed at interface